

Electrochemical Treatment of Perfluoroalkyl Acid (PFAA) Precursors and PFAAs in Groundwater Impacted with Aqueous Film Forming Foams

CDM Smith

Charles Schaefer, Ph.D.

Colorado School of Mines

Prof. Christopher Higgins

Prof. Timothy Strathmann

Andrew Maizel

Aniela Burant

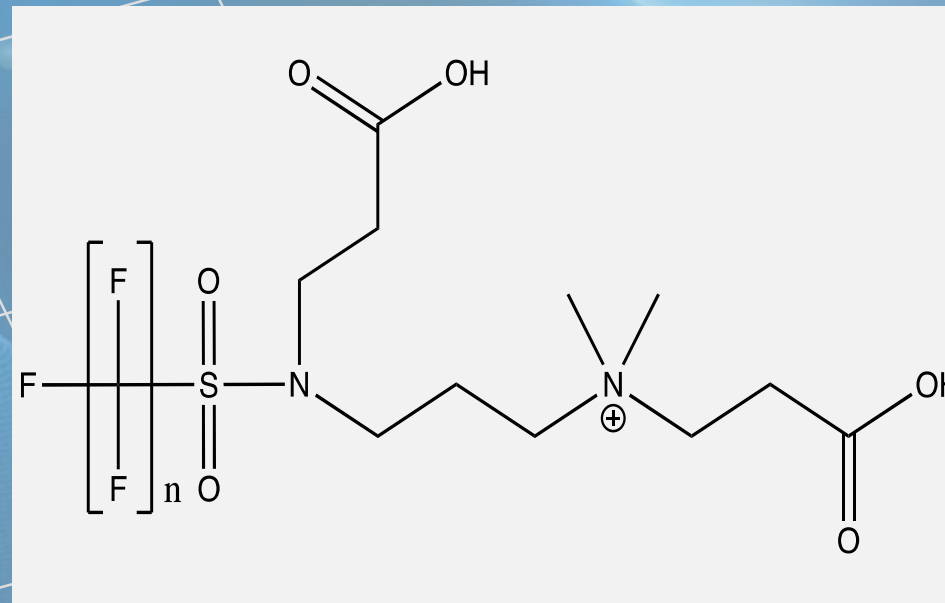
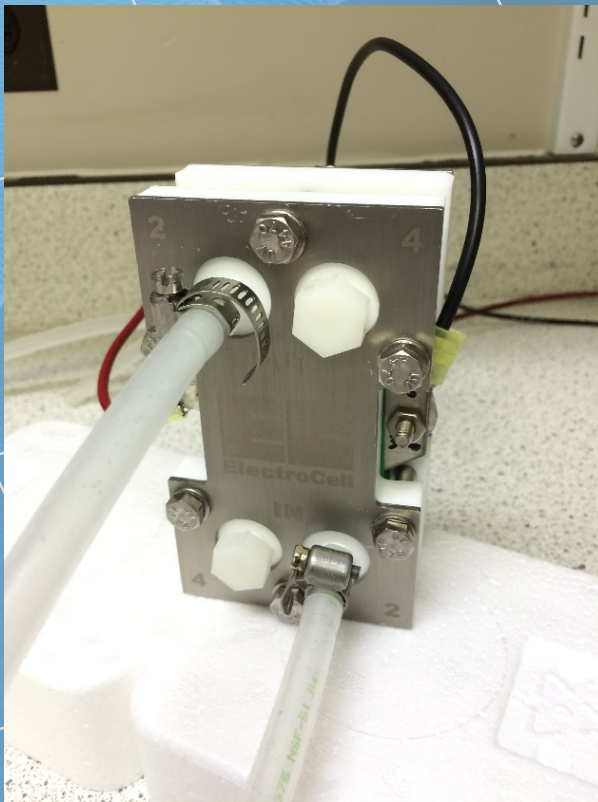
Duke University

Prof. Lee Ferguson

Sarah Choyke

APTIM

Christina Andaya



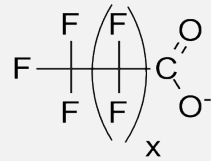
April 17, 2019

**CDM
Smith**[®]

Per- and Polyfluoroalkyl Substances (PFASs)

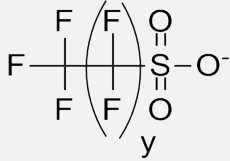
Perfluoroalkyl acids (PFAAs)

Carboxylates



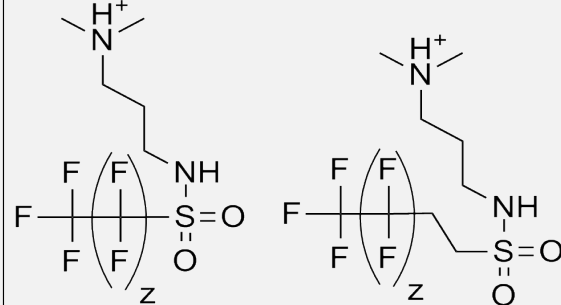
x = 6 PFOA

Sulfonates



y = 7 PFOS

PFAAs



PFAA Precursors

PFASs



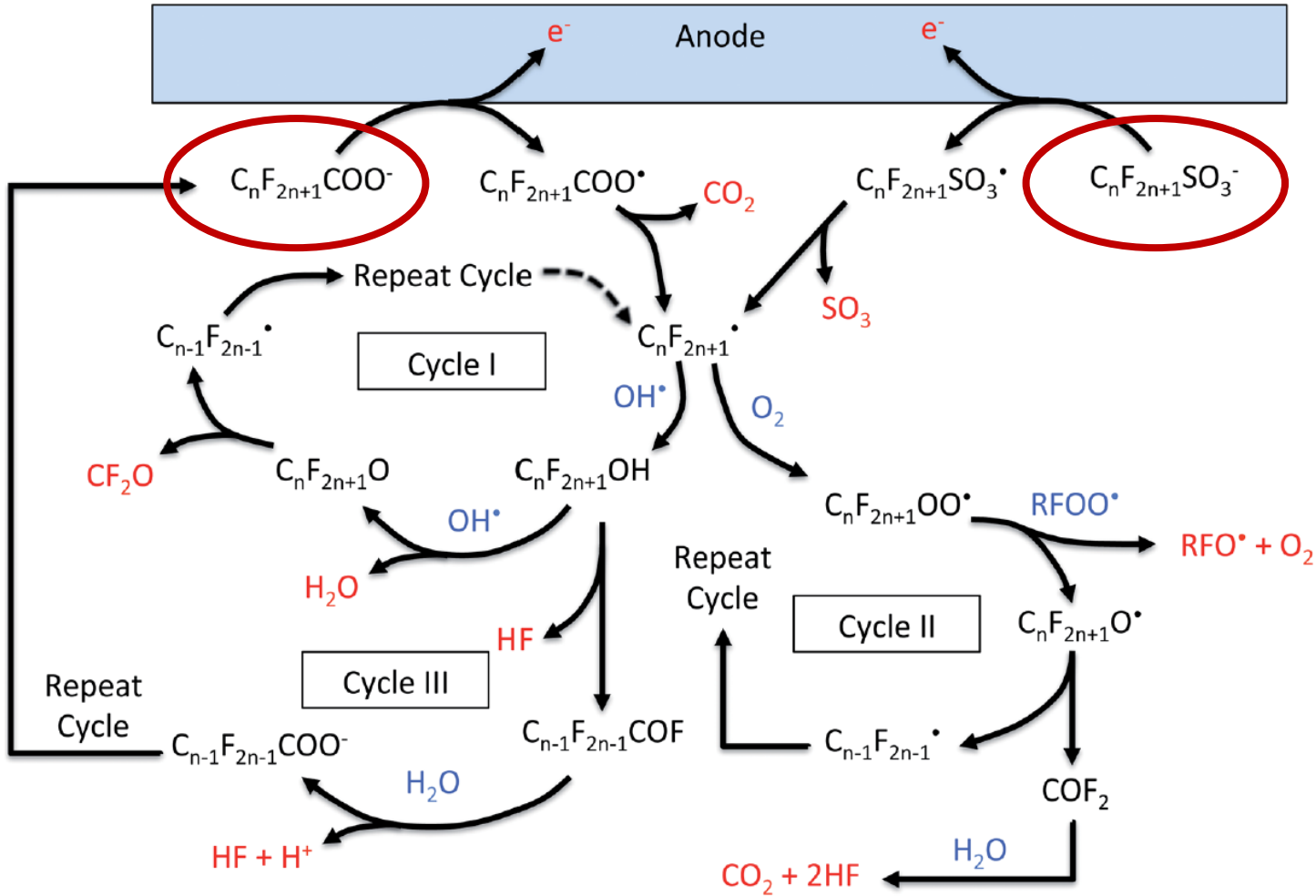
USEPA Health Advisory

70 ng/L (PFOS+PFOA)

Electrochemical Treatment of PFASs

- **Several studies evaluating EC treatment of PFASs (most for PFAAs)**
- **Several different anode materials**
 - Doped mixed metal oxide
 - Magnéli phase (Ti_4O_7)
 - Boron Doped Diamond
- **Perchlorate generation**
 - Biotic reduction
 - Chloride-free solutions
- **Areas for continued study**
 - Limited studies in real groundwater
 - Treatment for the wide range of PFASs present
 - Electrode longevity with respect to PFAS treatment
 - Energy consumption

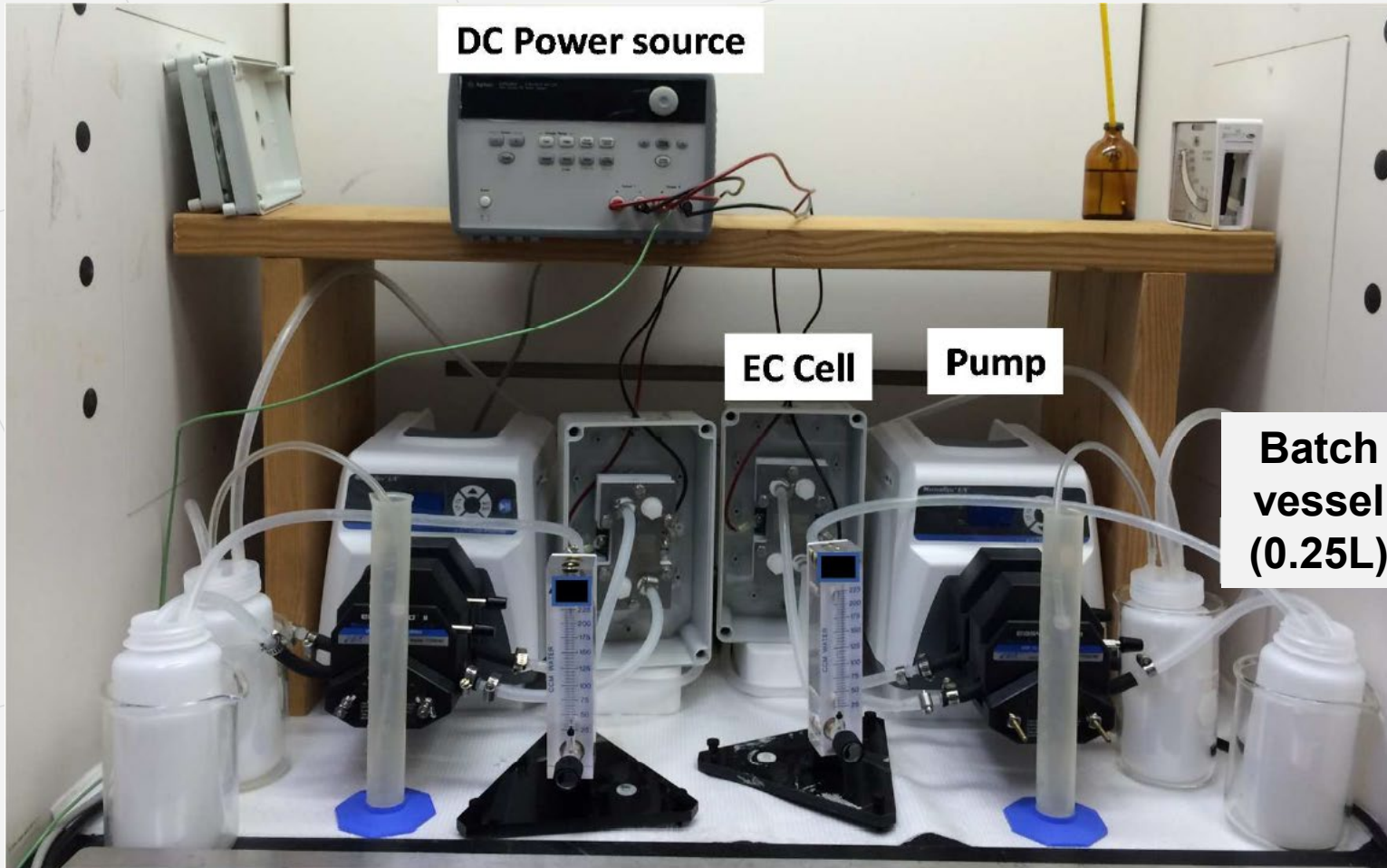
PFAA Electrochemical Oxidation Mechanism



- Transformation of both PFOA & PFOS begins with direct electron transfer at anode
- Functional groups removed, forming perfluoro radicals
- Unzipping process yielding shorter chain carboxylates and fluoride *(not always observed)*

figure from: Chaplin, *Environ. Sci.: Processes*, 2014

Experimental



8 hour batch experiments
25 mA/cm² current density

BDD Anode (Condias)

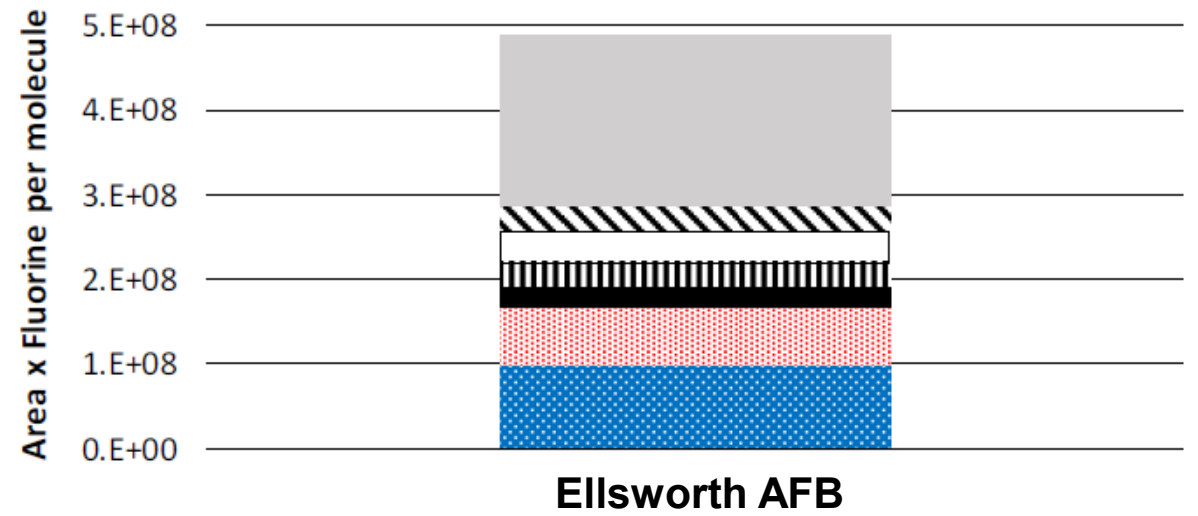
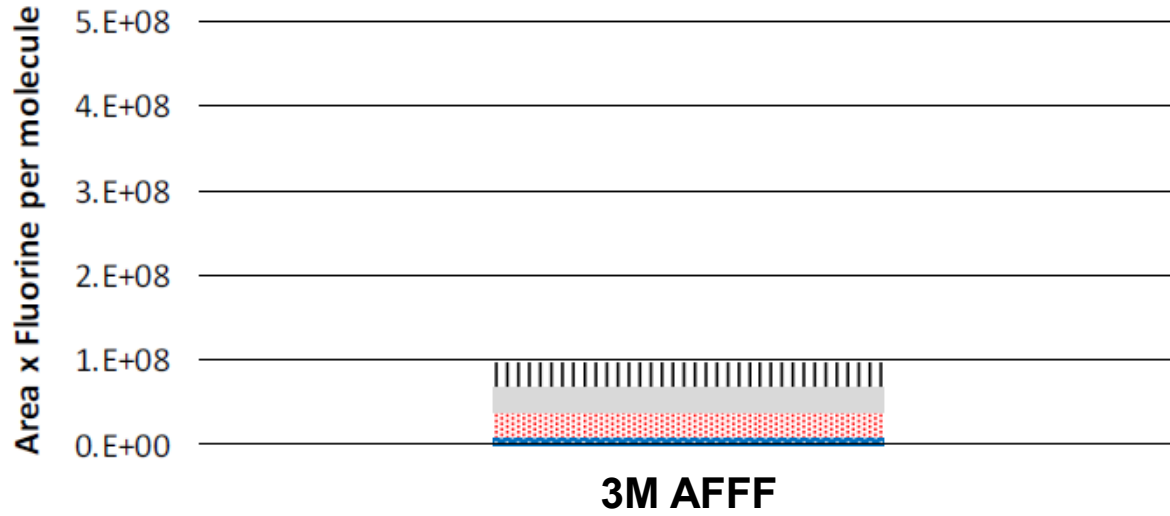
- BDD on Nb
 - SS cathode
 - 4mm spacing
- Natural groundwater (spiked with 3M AFFF)
 - AFFF-impacted groundwater (Ellsworth AFB)
 - LC-MS/MS and HR-MS
 - F mass balance
 - Long-term experiment

3M AFFF in GW**Ellsworth AFB GW**

Fluoride (mg/L)	<0.2	0.9
Chloride (mg/L)	18	160
Sulfate (mg/L as SO₄)	25	14
Nitrate (mg/L as N)	5.5	<0.2
Total Organic Carbon (mg/L)	11	5.1
Turbidity (NTU)	1.8	6.3
Alkalinity (mg/L CaCO₃)	352	890
Conductivity (µmhos/cm)	769	2120
pH	6.7	7.1
Hardness (mg/L CaCO₃)	363	1,030

	3M AFFF in GW	Ellsworth AFB GW
<i>PFAAs (µg/L)</i>		
PFBA	<0.074	25
PFPeA	4.0	61
PFHxA	9.0	130
PFHpA	2.6	12
PFOA	15	58
PFBS	33	45
PFHxS	76	160
PFHpS	11	26
PFOS	300	22

Potential PFAA Precursors



■ CEtAmPr-FASA-PrA
 ■ AmPr-FASA-PrA
 ■ AmPr-FASA || FASA

■ S-OHPrAmPr-FASA-OHPrS

■ SPrAmPr-FASAPrS

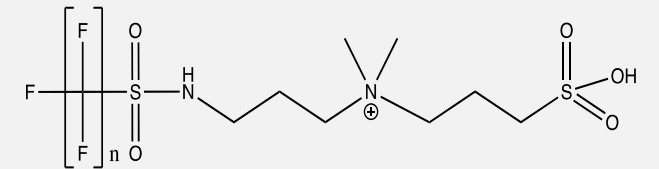
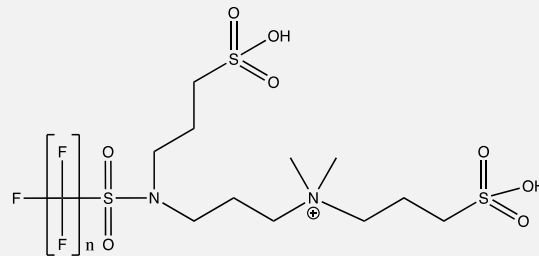
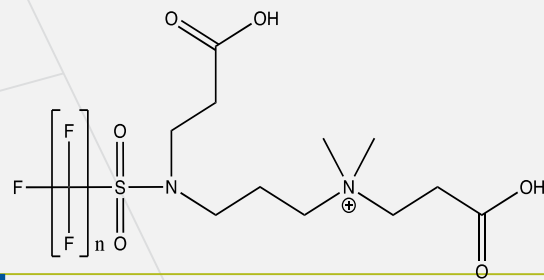
■ S-OHPrAmPrFASAA

■ SPrAmPr-FASAA

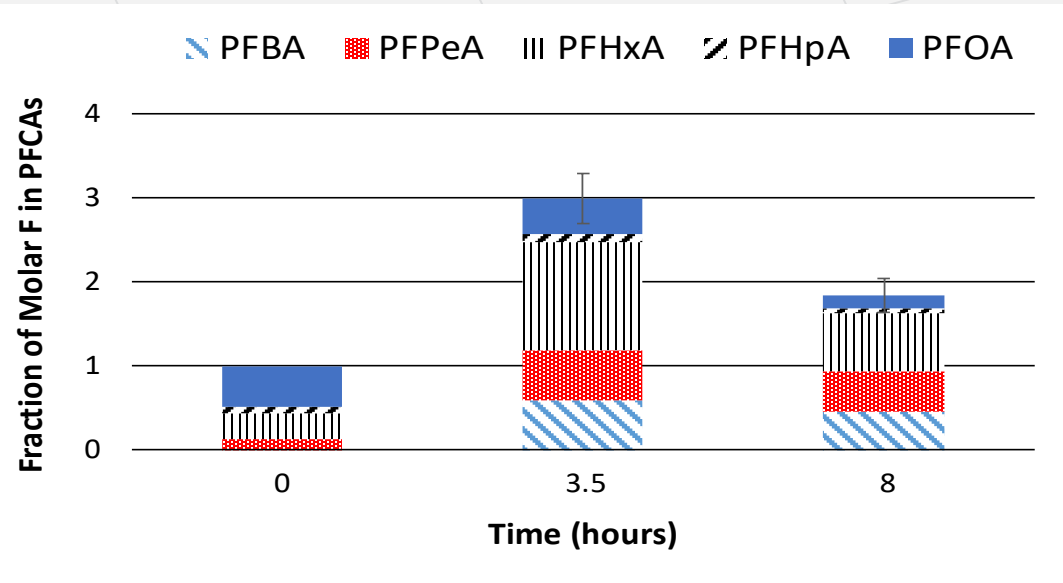
■ SPrAmPr-FASA

■ SPr-FASA

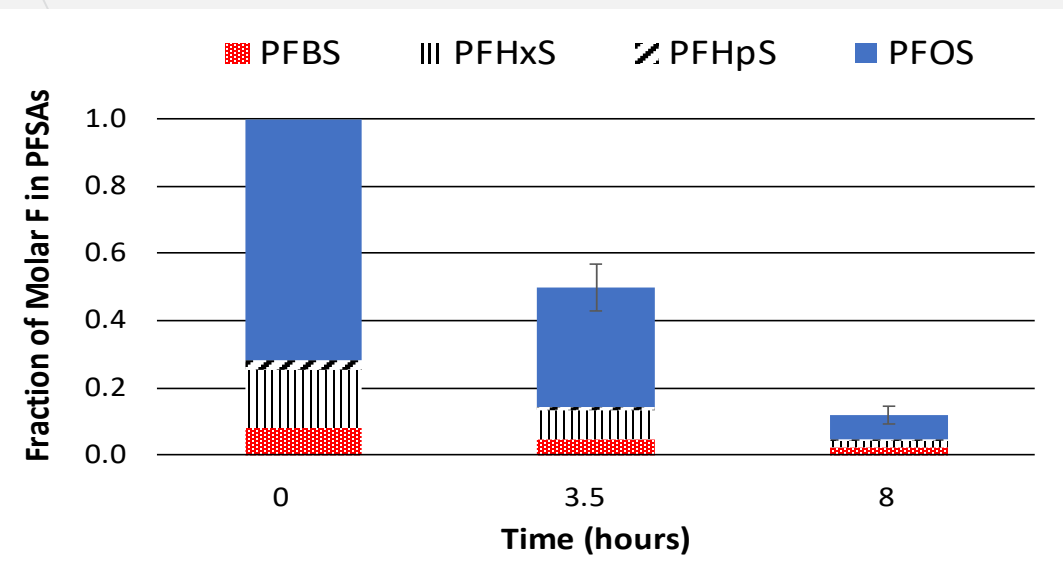
■ AmPr-FASA



EC Treatment (3M AFFF)



PFCAs show transient increases, but no increases in PFSAs

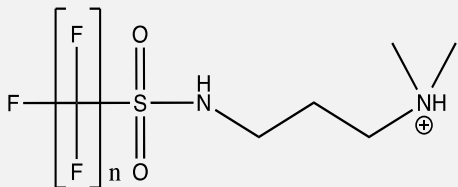
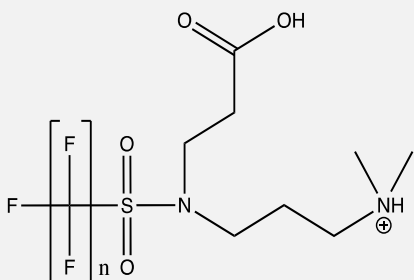
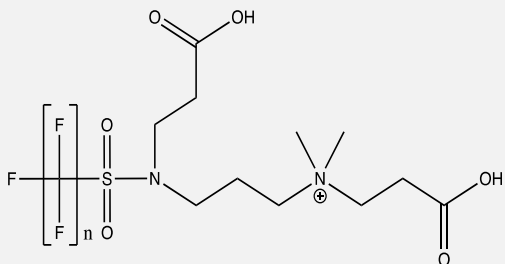


Total Oxidizable Precursor assay
Showed substantial PFCAs generation for untreated 3M AFFF-spiked water

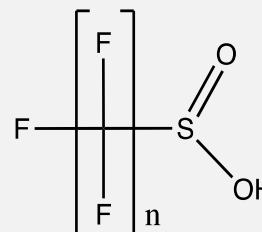
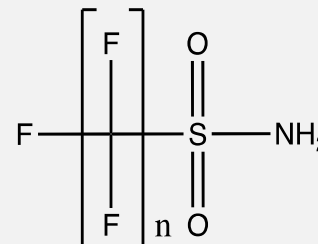
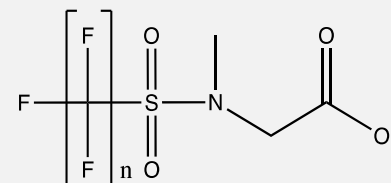
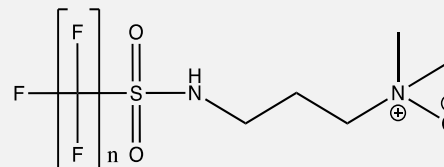
Identified Pre-Cursors in 3M AFFF Spiked Groundwater

HR-MS data: semi-quantitative

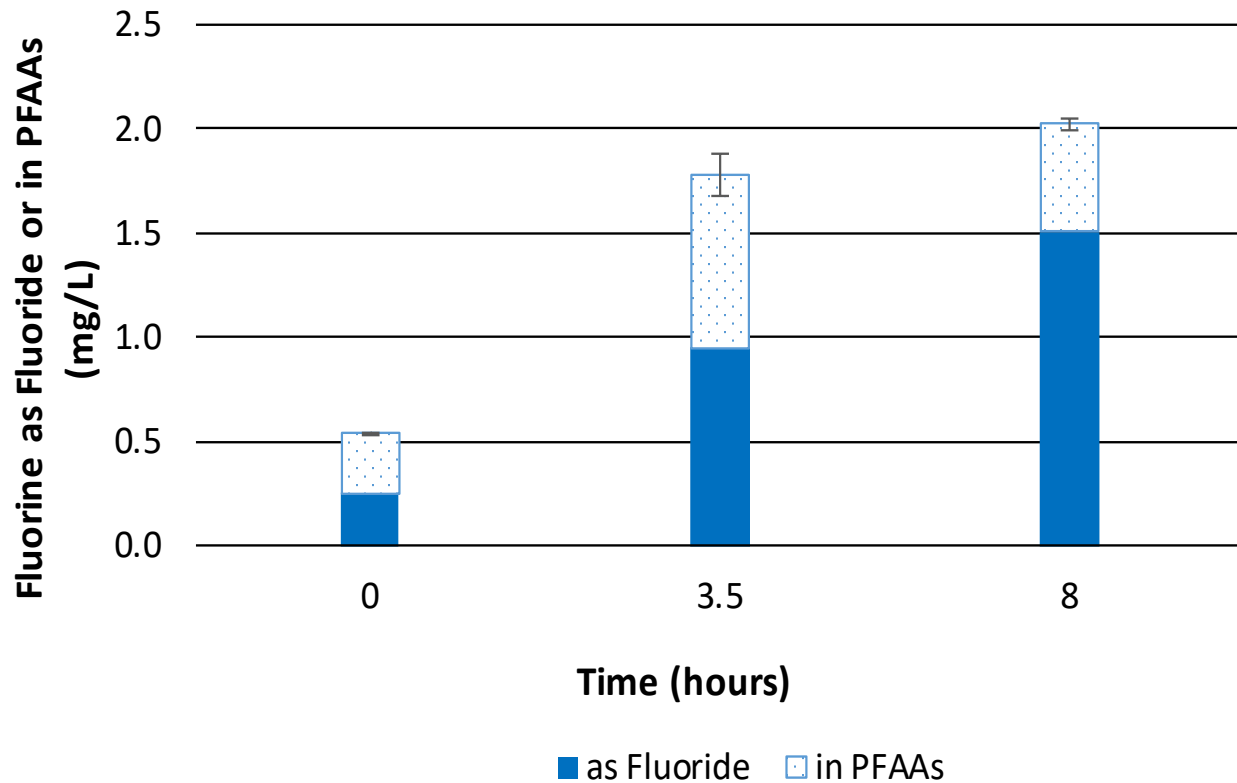
Rapidly Removed



Transient Increase



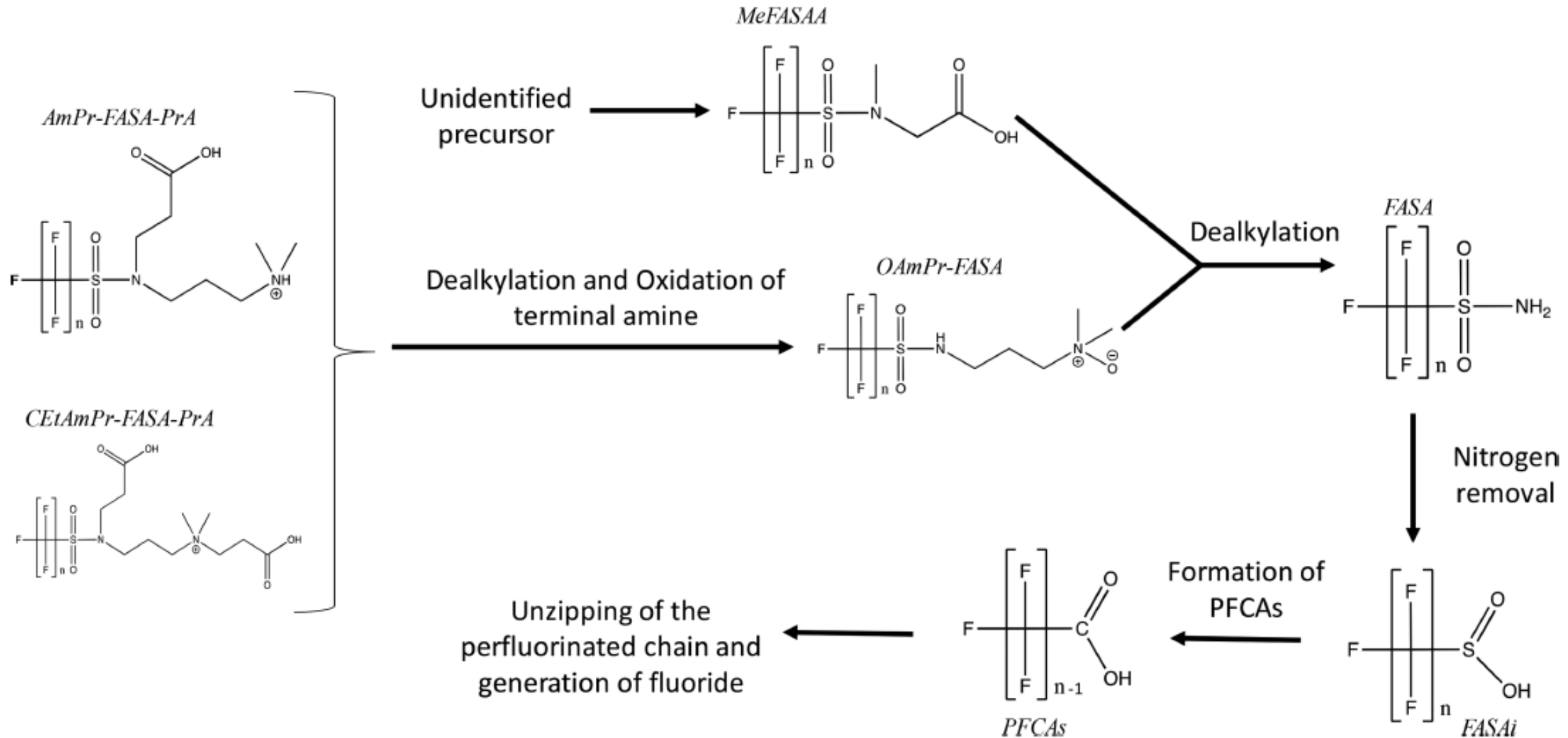
F Mass Balance (3M AFFF)



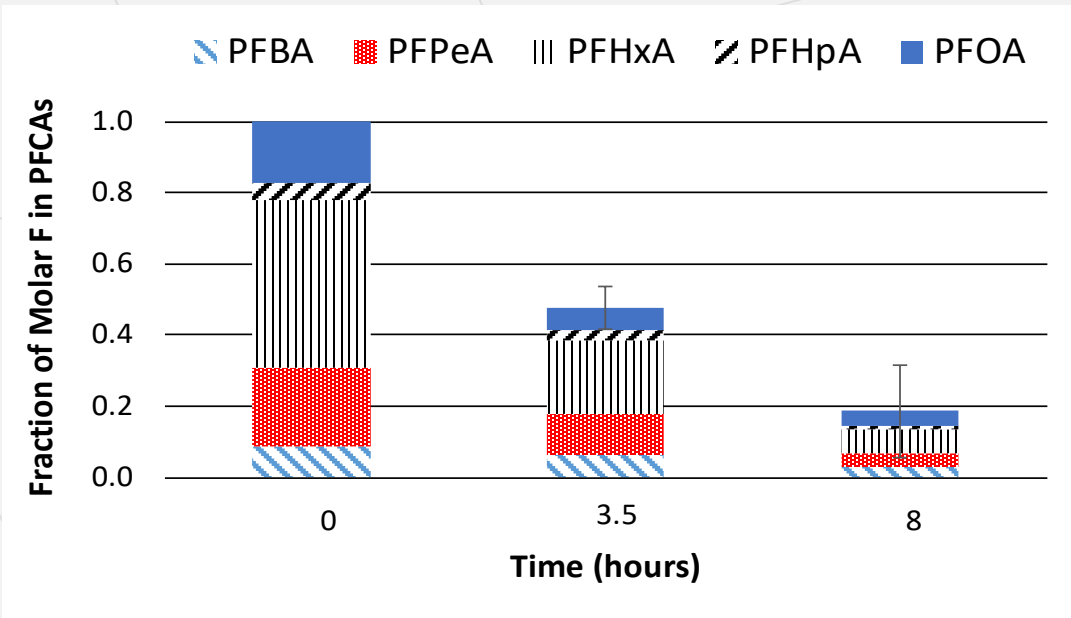
PFAAs in untreated water could only account for about 1/3 of the fluoride generated

Based on TOP assay, 64% of the F present as PFAS accounted for

Oxidation Pathway during EC Treatment: 3M AFFF

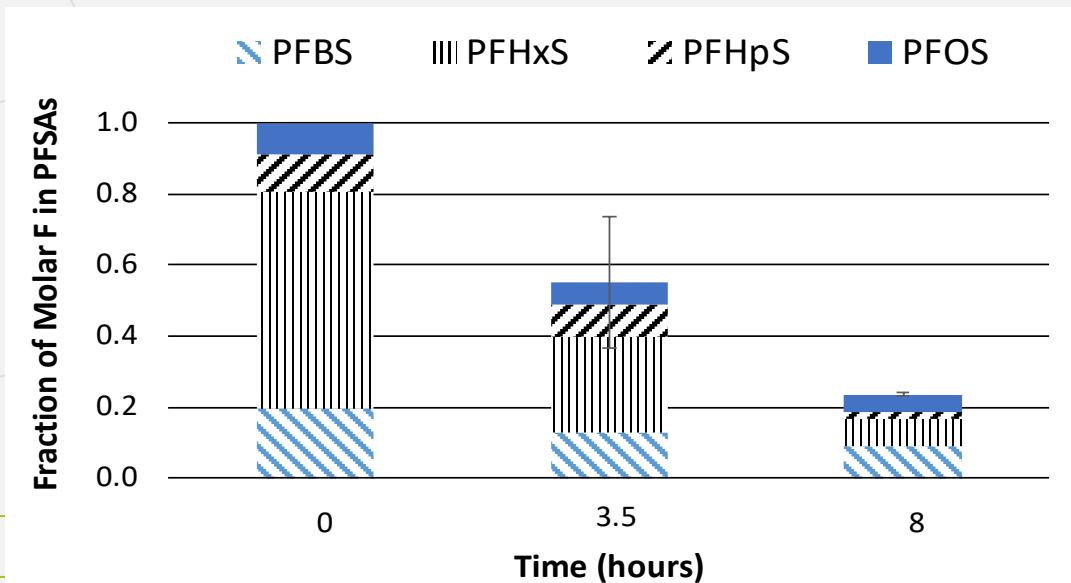


EC Treatment (Ellsworth AFB GW)



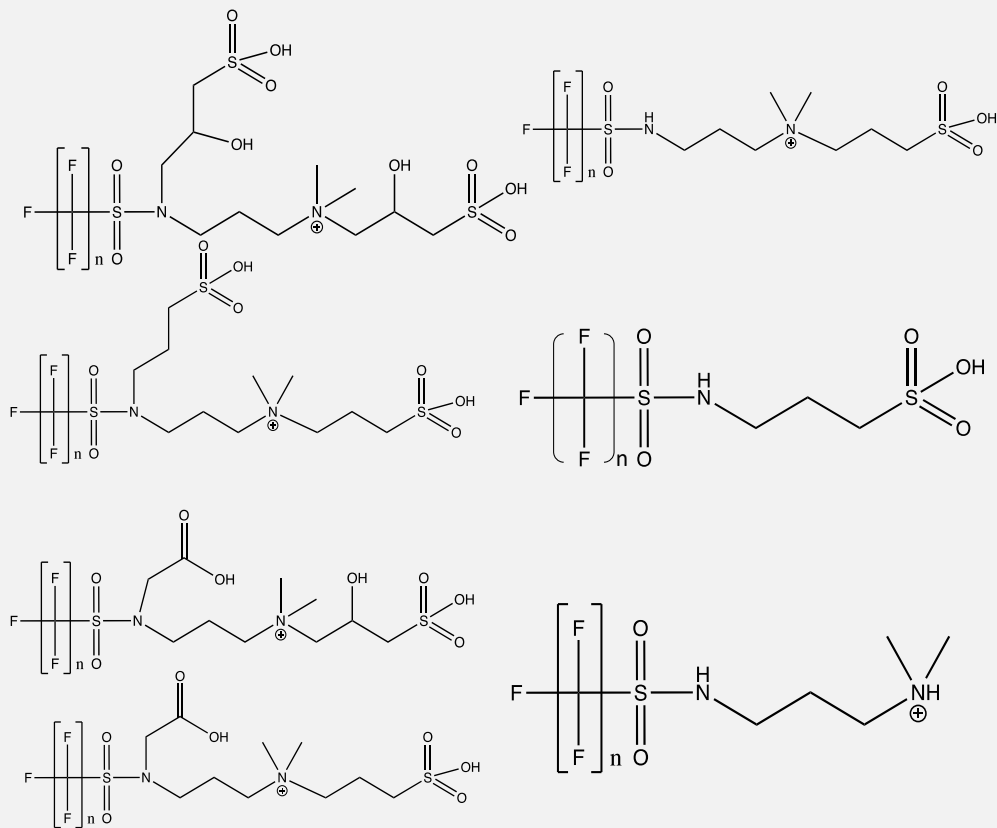
***NO transient increases
in PFCAs or PFSA***

Total Oxidizable Precursor assay
No PFCA or PFSA generation for
untreated Ellsworth AFB groundwater



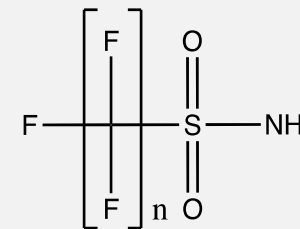
Fate of Identified Pre-Cursors in Ellsworth AFB Groundwater during EC Treatment

Rapidly Removed

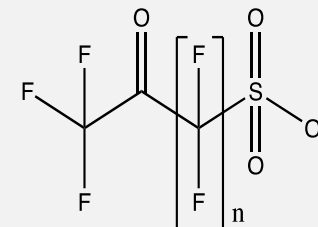


Transient Increase

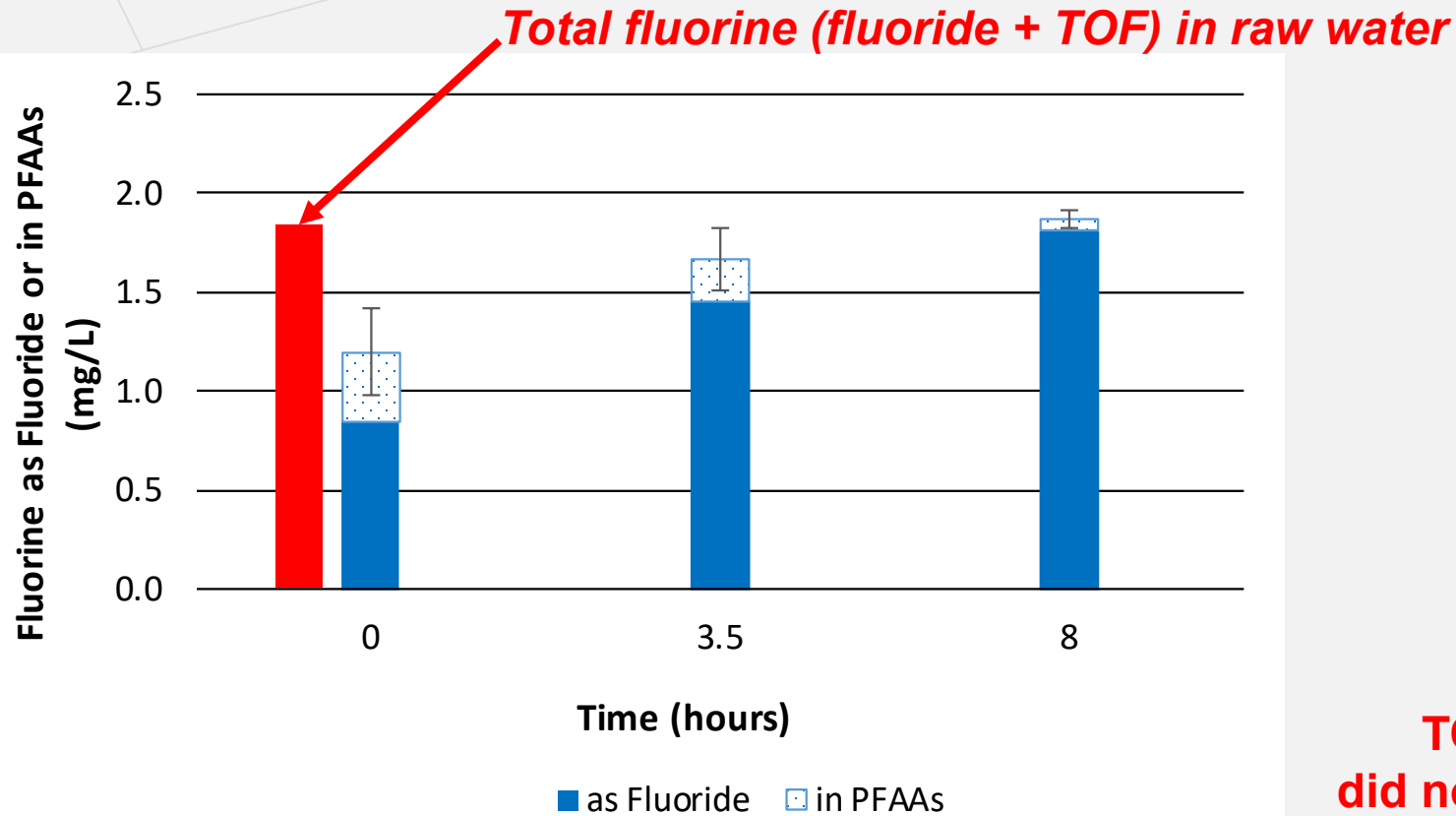
10 to 100x
Less than in 3M
AFFF GW



Not observed
in 3M AFFF GW



F Mass Balance (Ellsworth AFB Groundwater)



Only 1/3 of the fluoride generated could be accounted for by the PFAAs initially present

TOP analysis or standard PFAS analysis did not capture ~65% of organic fluorine present

Electrochemical Efficiency

Coulombic Efficiency for Defluorination

$$CE = FV_e \frac{C_F}{MW_{FA}t} \sim 2 \times 10^{-4}$$

3M AFFF-Spiked GW

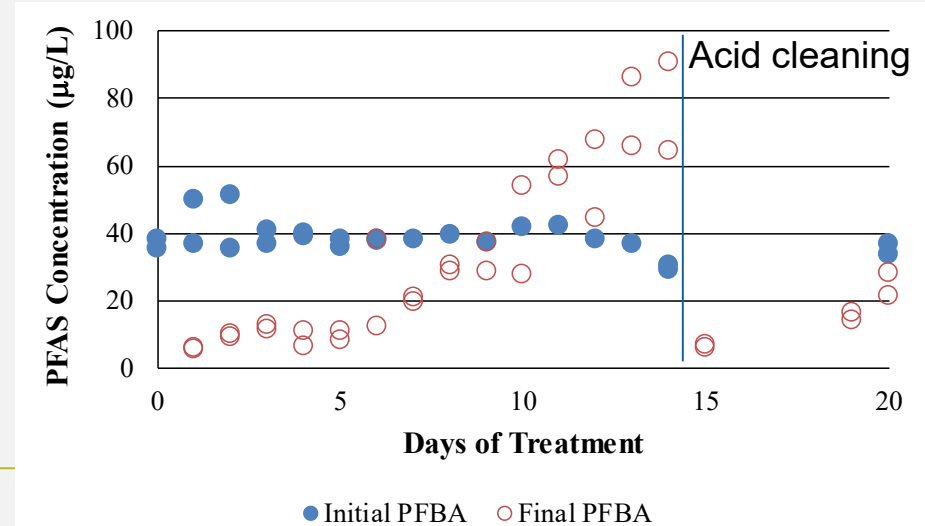
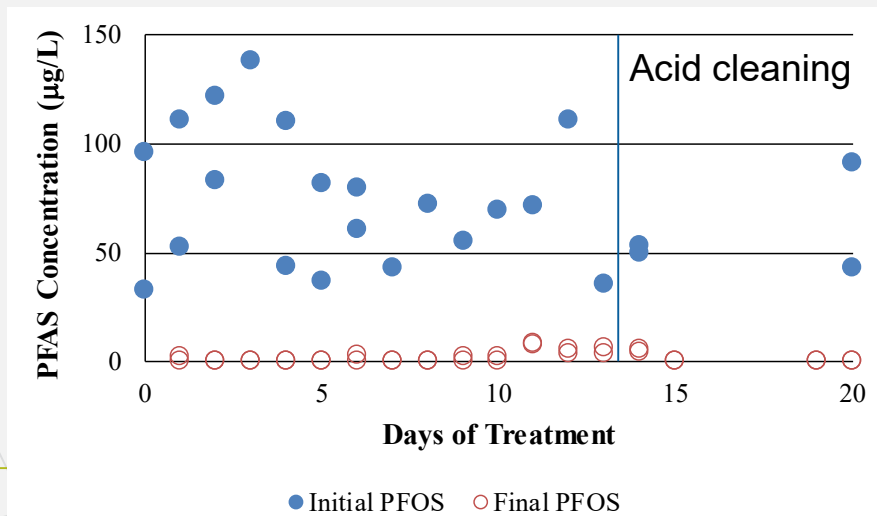
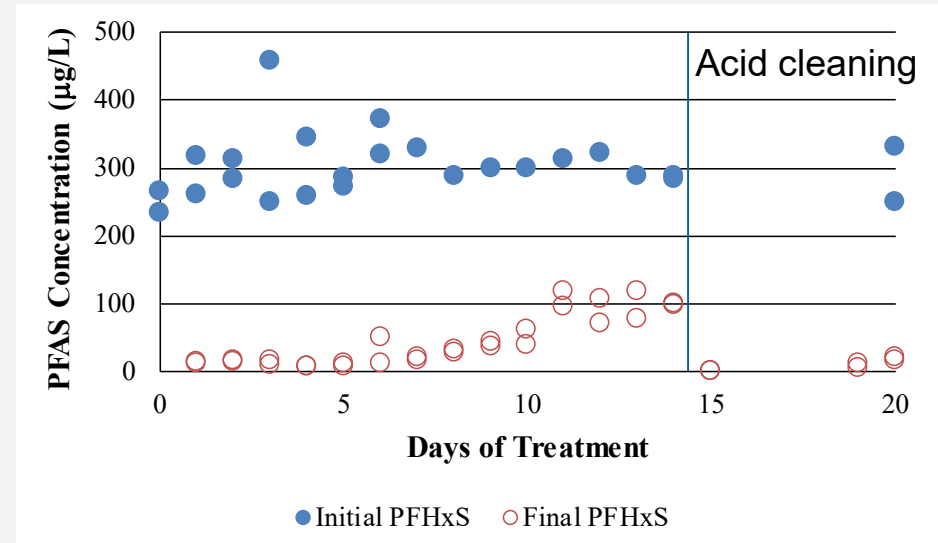
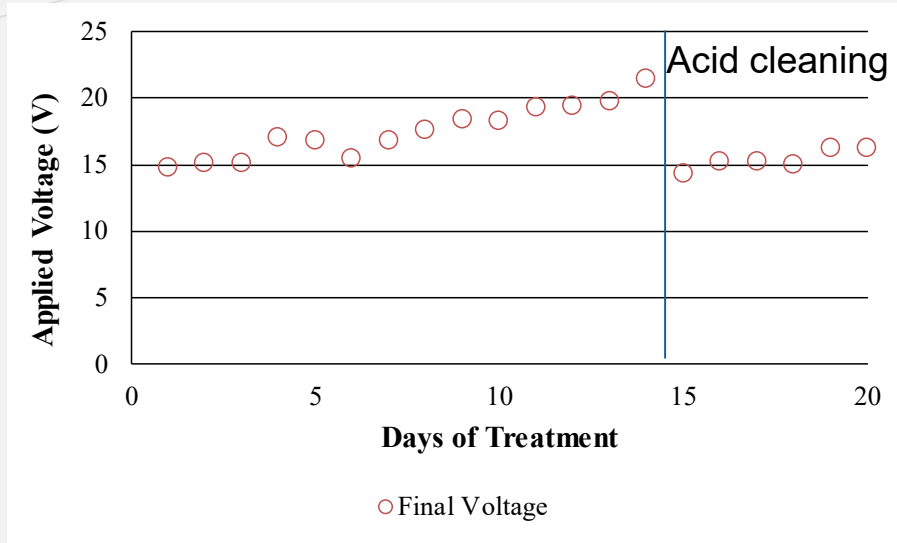
<u>PFAS</u>	<u>% Removal</u>	<u>Energy (W-h/L)</u>
PFOS	91	
PFHpS	94	
PFHxS	87	
PFBS	78	99
PFOA	71	
PFHpA	45	
PFHxA	(increase)	
PFPeA	(increase)	
PFBA	(increase)	

Ellsworth AFB GW

<u>PFAS</u>	<u>% Removal</u>	<u>Energy (W-h/L)</u>
PFOS	40	
PFHpS	85	
PFHxS	89	
PFBS	63	136
PFOA	75	
PFHpA	59	
PFHxA	81	
PFPeA	67	
PFBA	57	

Long Term Testing

(Successive 24-hour batch tests using Ellsworth AFB Groundwater)



Summary

- EC treatment remains a promising technology for PFAS, esp. for high PFAS concentrations
- Poly fluorinated compounds, and their oxidation pathways, are important and not yet fully understood
- The TOP assay may not be a good indicator of polyfluorinated compounds present in water. Total organic fluorine via CIC may be a better indicator.

Research Funding

