

Evidence for Anaerobic Microbial Dechlorination of Chlordecone in Microcosms and Soil from Guadeloupe

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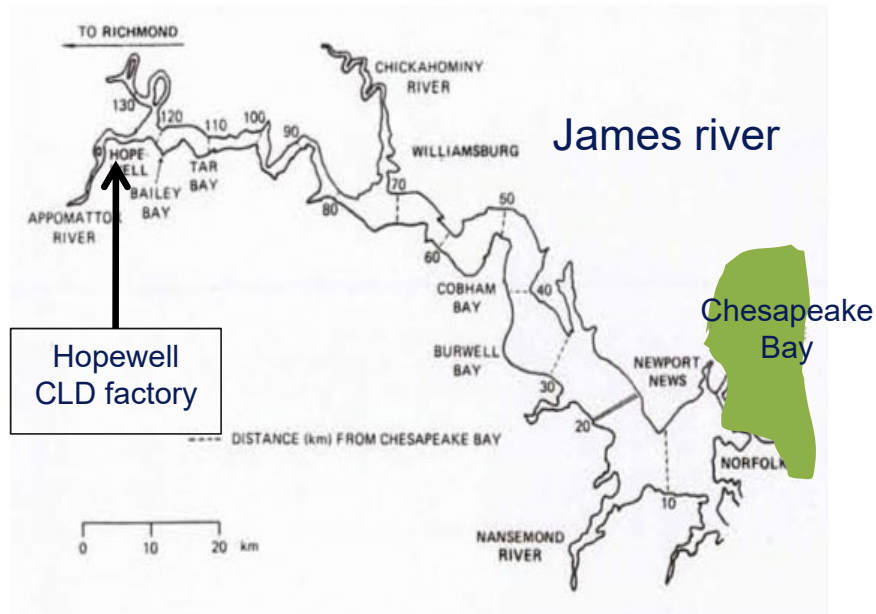
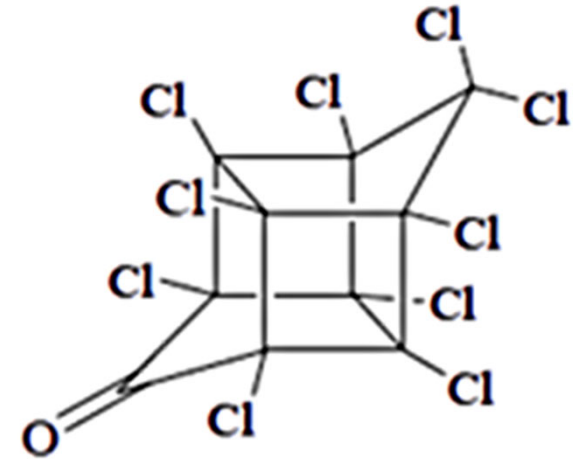
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History of Chlordecone

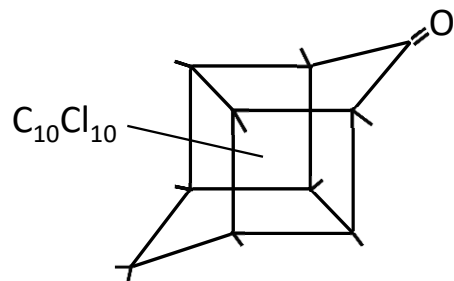
- 1951: First synthesis; 1952 US Patents
- 1958: First commercialized by Allied Chemical Company under the name Kepone[®] / produced in USA: Delaware, Pennsylvania, Virginia (Hopewell)



- **1966–1975** - The Allied Chemical Corporation dumps Kepone, a toxic, nonbiodegradable insecticide, into Virginia's James River. Its effect on the environment is eventually publicized, leading authorities to shut down the Allied Chemical Corporation plant that produced the chemical and to order fishing bans and advisories.

Chlordecone

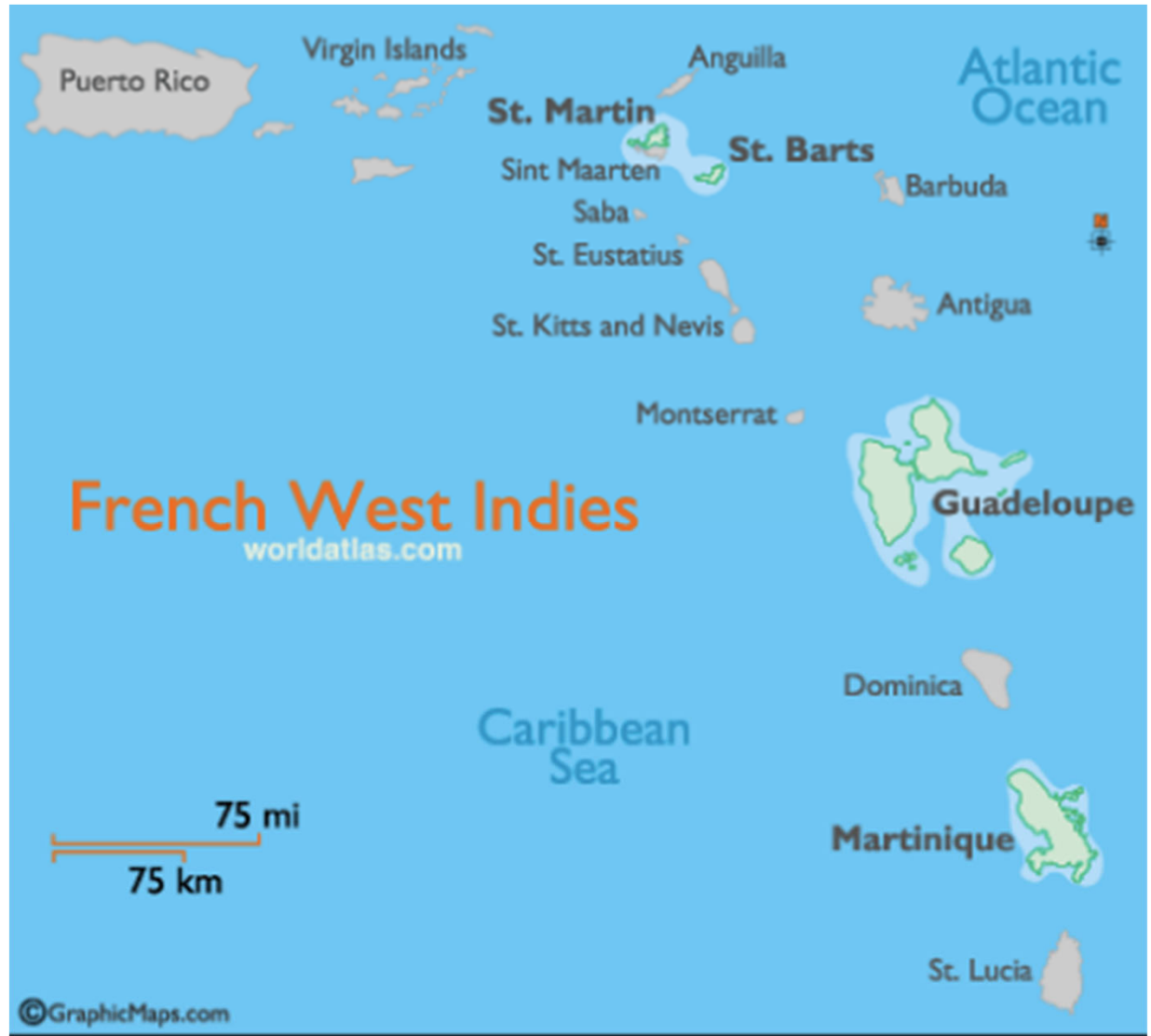
- Pesticide that was extensively used in banana and sugar cane plantations in the Carribbean between 1972 and 1993
- Has led to the pollution of large agricultural areas
- Recalcitrant and adsorbs to soil
- Carcinogen



Chlordecone (kepone)
 $C_{10}Cl_{10}O$

Chemistry suggests that anaerobic dechlorination is possible, but few studies have shown that biological dechlorination is happening in the environment





Objective: Investigate anaerobic biotransformation of chlordecone in microcosms



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Microcosm setup December 2010

Development of analytical procedures

2018: Analysis of new field samples from Guadeloupe

Microcosm setup
(Laurent Laquitaine)

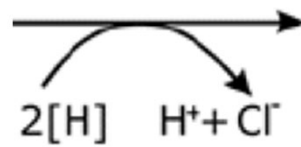
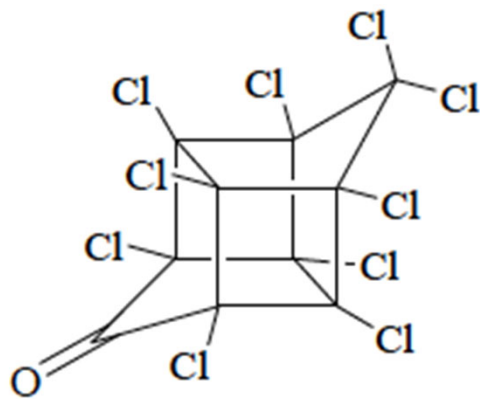


Field sampling

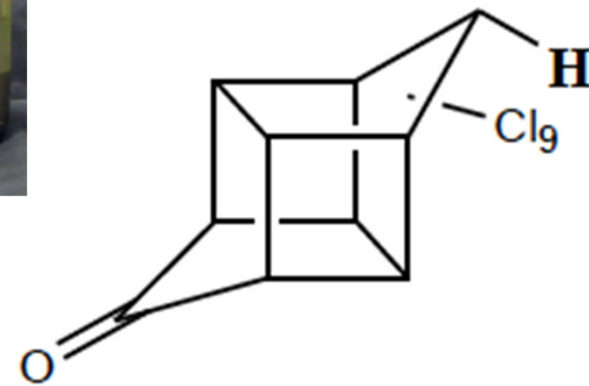


Chlordecone Microcosms

Chlordecone: $C_{10}Cl_{10}O$



Hydrogenolysis



Monohydrochlordecone:
 $C_{10}HCl_9O$

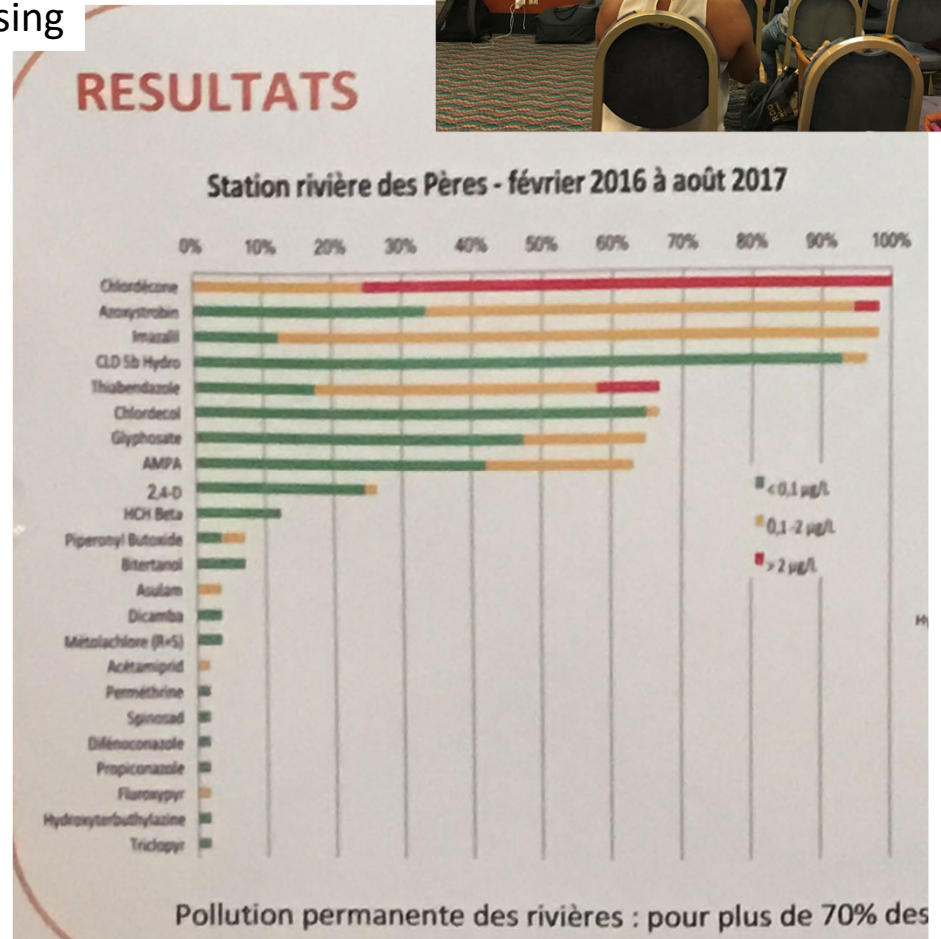


Line Lomheim

Sarra Gaspard

Suly Rambinaising

October 2018; Guadeloupe/Martinique

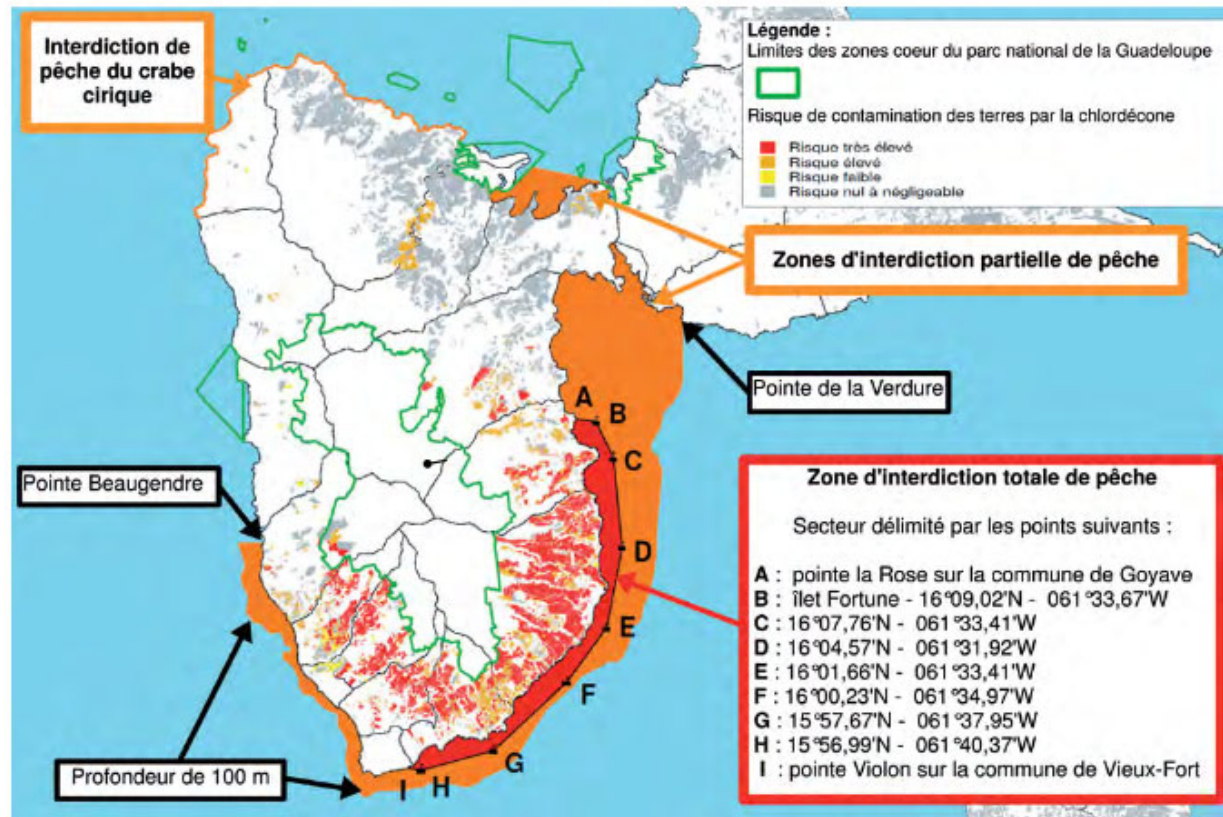


Impact on the Economy

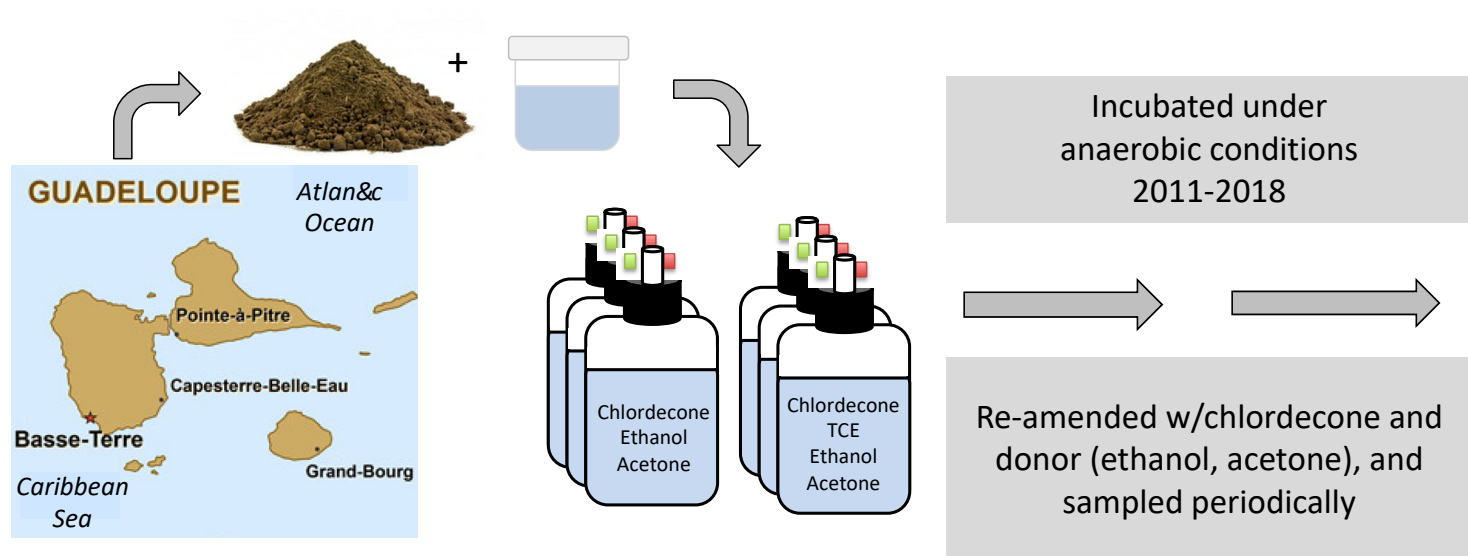
Restriction of use of contaminated soils

Fishing Restrictions

Restrictions on fish farming

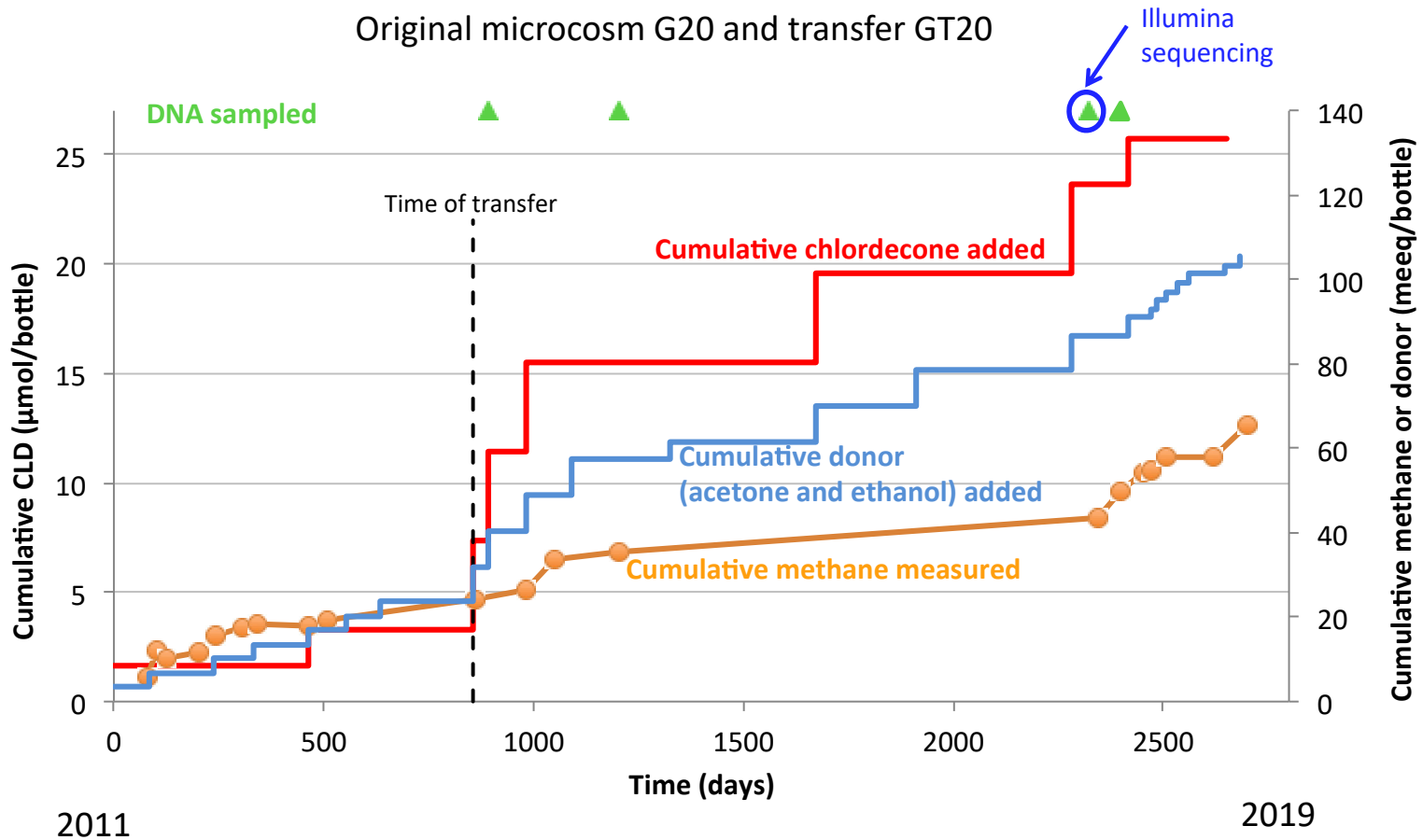


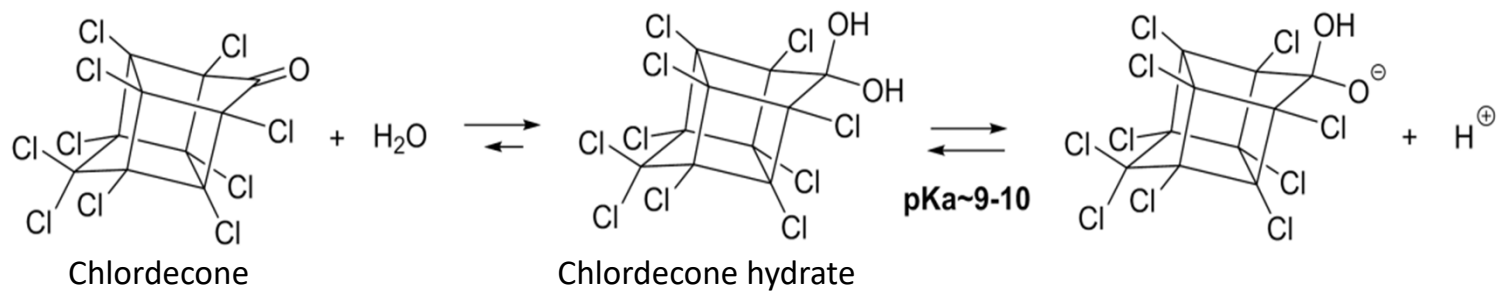
Construction of anaerobic microcosm from Guadeloupe soil



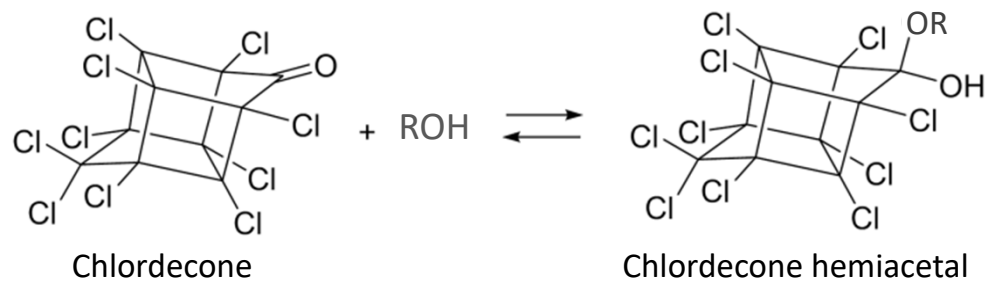
Overview of Microcosms

Original microcosm G20 and transfer GT20





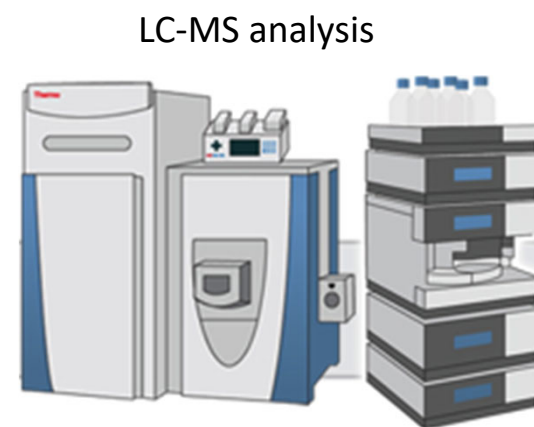
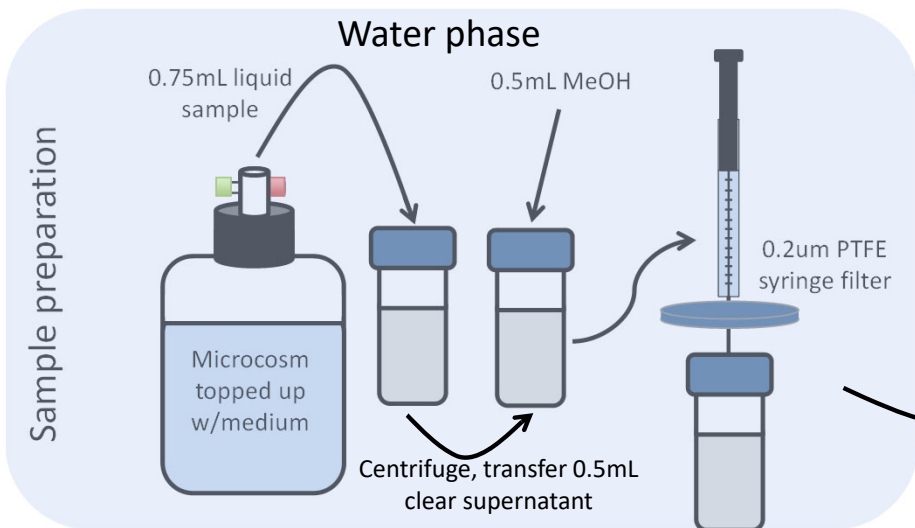
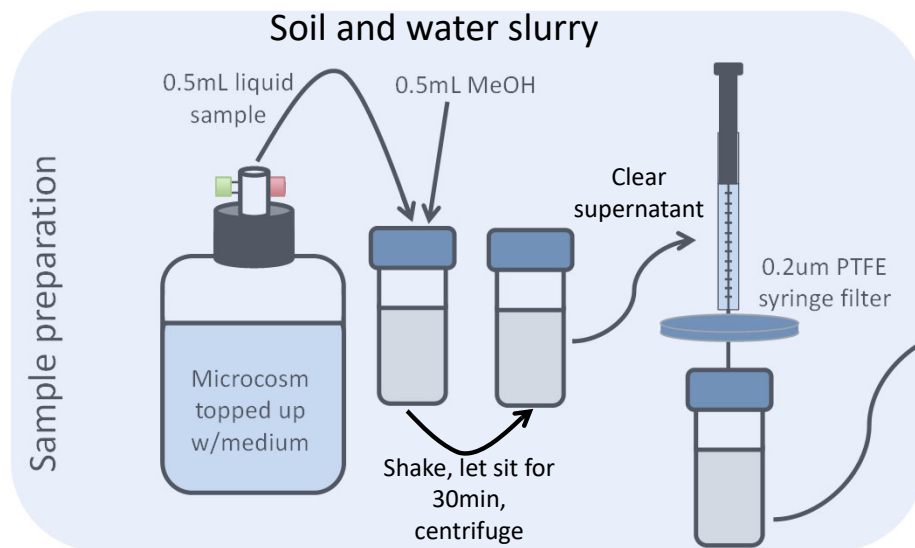
The carbonyl or keto group (C=O) can combine with water to form a geminal diol



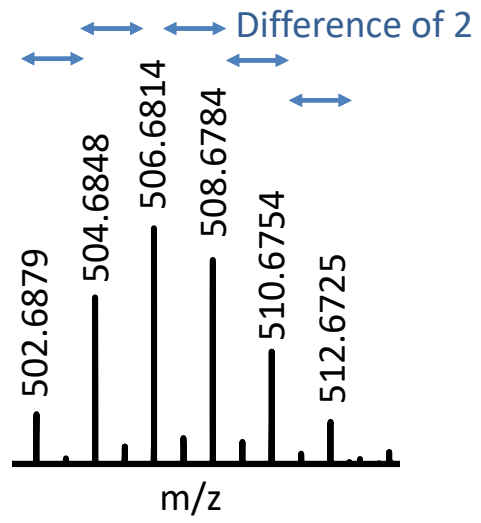
Chlordecone in the presence of alcohol (R = methanol or ethanol)

The alcohol is added to the ketone to form the hemiacetal (or hemiketal)

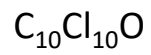
Analysis of chlordecone and degradation products



Q-Exactive Orbitrap mass spectrometer
w/electrospray ion source
(Thermo Fisher) in
negative ion mode

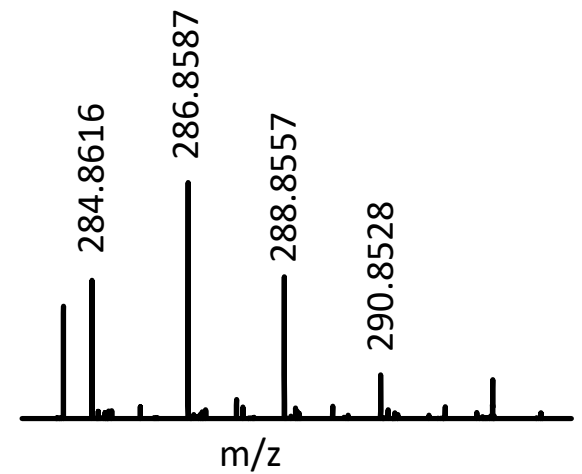


Chlordecone

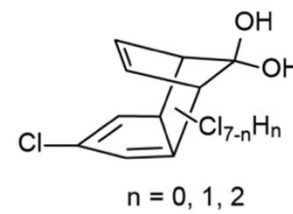
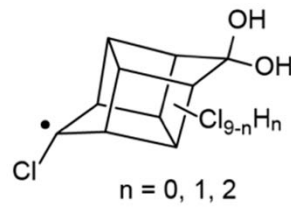
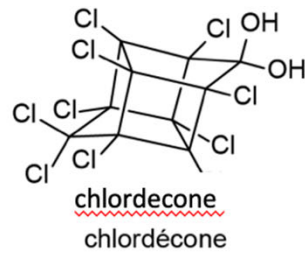


Cl-35 - 75.8%

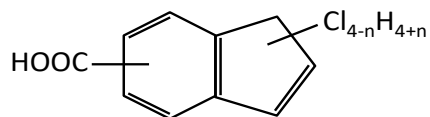
Cl-37 - 24.2%



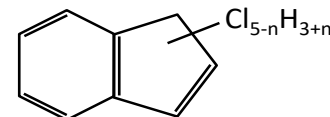
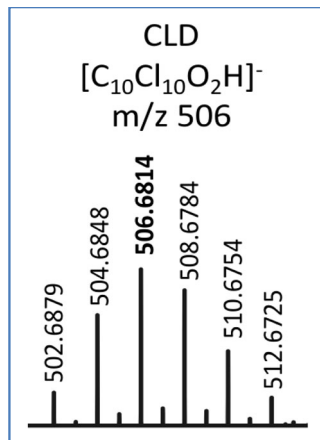
Must be a metabolite



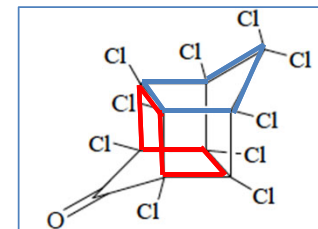
Mass Spectra



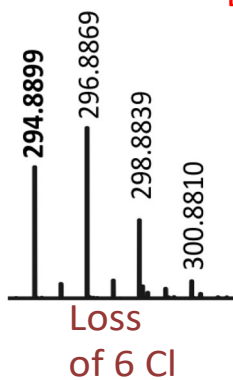
Carboxylated polychloroindenes
(n=0,1,2,3)



Polychloroindenes
(n=0,1,2)

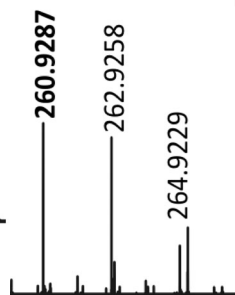


C1-C2
[C₁₀Cl₄O₂H₃]⁻
m/z 294



Loss
of 6 Cl

C3-C7
[C₁₀Cl₃O₂H₄]⁻
m/z 260



Loss
of 7 Cl

C8-C11
[C₁₀Cl₂O₂H₅]⁻
m/z 226



Loss
of 8 Cl

C12-C13
[C₁₀ClO₂H₅]⁻
m/z 193



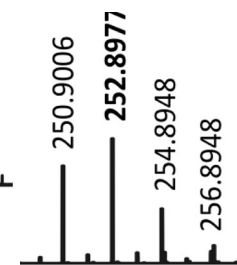
Loss
of 9 Cl

B1
[C₉Cl₅H₂]⁻
m/z 286



Loss
of 5 Cl

B2
[C₉Cl₄H₃]⁻
m/z 252



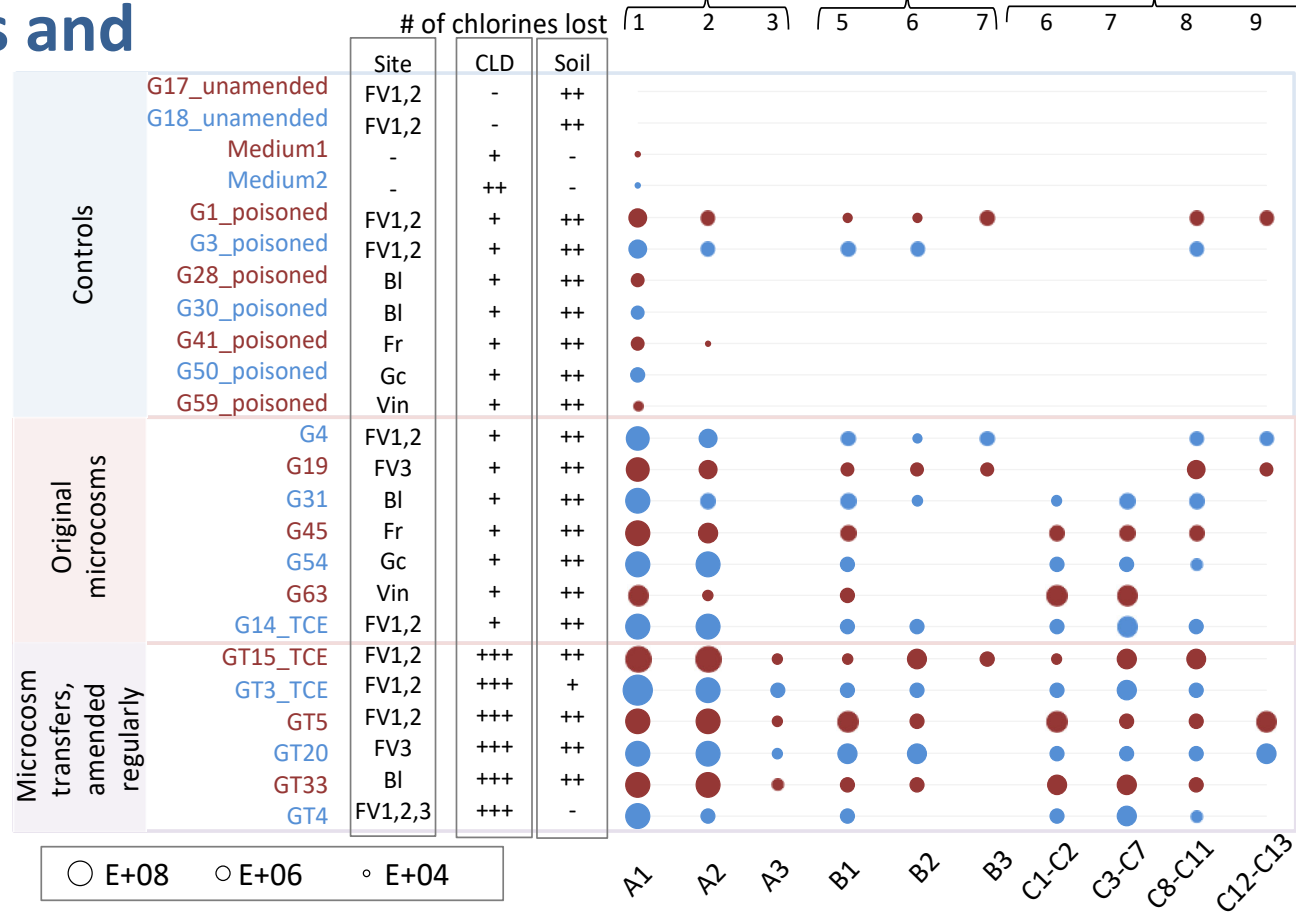
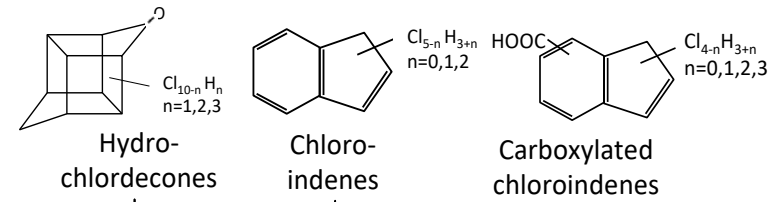
Loss
of 6 Cl

B3
[C₉Cl₃H₄]⁻
m/z 252

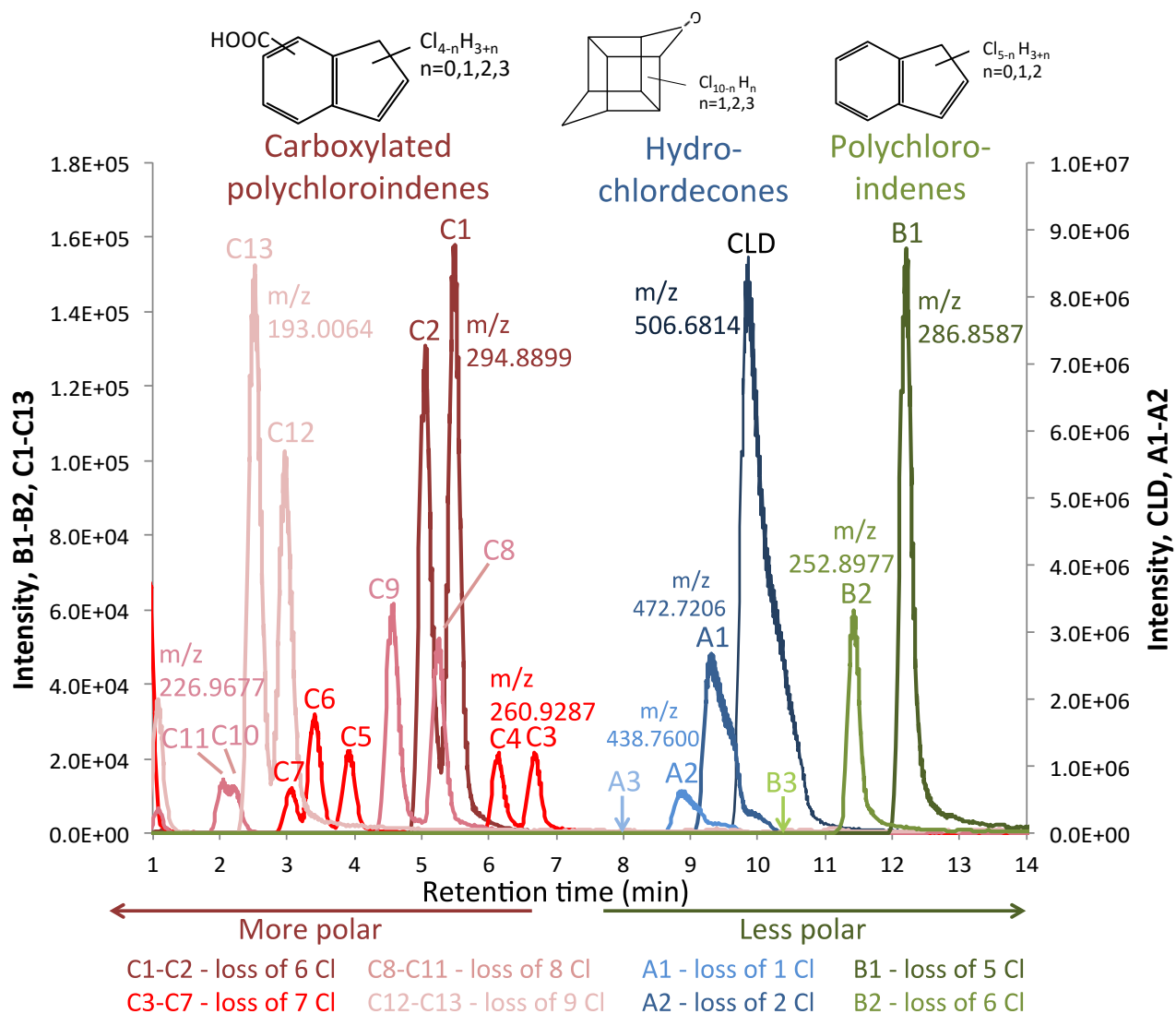


Loss
of 7 Cl

Degradation products observed in microcosms and controls



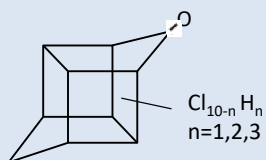
Chromatogram of chlordecone and its dechlorination products



Sample G20_soil from June 29th 2018

Structure and formula of dechlorination products observed in the anaerobic microcosms

Reductive dechlorination products (loss of 1-3 chlorines)

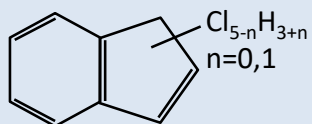


$\text{C}_{10}\text{Cl}_9\text{OH}$ Monohydrochlordecone (**A1**)

$\text{C}_{10}\text{Cl}_8\text{OH}_2$ Dihydrochlordecone (**A2**)

$\text{C}_{10}\text{Cl}_7\text{OH}_3$ Trihydrochlordecone (**A3**)

“Open Cage” structures (loss of 5-9 chlorines)

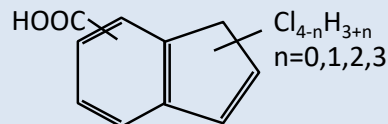


Chloroindenes

$\text{C}_9\text{Cl}_5\text{H}_3$ (**B1**)

$\text{C}_9\text{Cl}_4\text{H}_4$ (**B2**)

$\text{C}_9\text{Cl}_3\text{H}_5$ (**B3**)



Carboxylated chloroindenes

$\text{C}_{10}\text{Cl}_4\text{O}_2\text{H}_4$ (**C1-C2**)

$\text{C}_{10}\text{Cl}_3\text{O}_2\text{H}_5$ (**C3-C7**)

$\text{C}_{10}\text{Cl}_2\text{O}_2\text{H}_6$ (**C8-C11**)

$\text{C}_{10}\text{ClO}_2\text{H}_7$ (**C12-C13**)

1. Schrauzer, G. N., & Katz, R. N. (1978). *Bioinorg Chem*, 9(2), 123-143.
2. Ranguin, R., et al. (2017). *Environ Sci Pollut Res Int*. doi: 10.1007/s11356-017-9542-z.
3. Chaussonnerie, S., et al. (2016). *Front Microbiol*, 7(2025), 2025.

Proposed degradation pathways for CLD

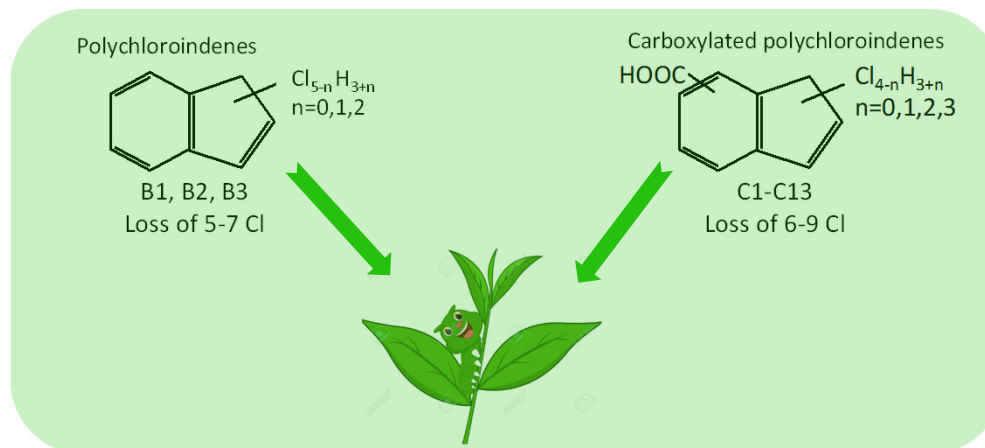
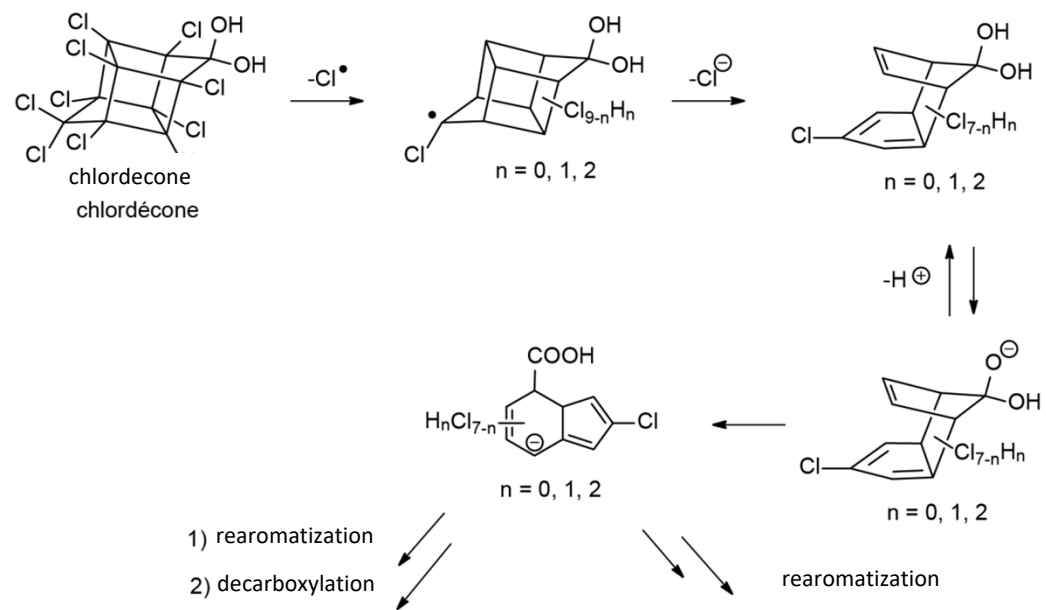


Figure from PhD thesis of Marion Chevallier: 2018

Etude de la dégradation biologique et chimique d'un pesticide persistant : la chlordécone

Thèse de doctorat de l'Université Paris-Saclay

Analysis of CLD and dechlorination products in microcosms

	Microcosm ID	CLD added to date	CLD remaining after 8 years	CLD Recovered	Sum MHCLD, DHCLD and THCLD*	Sum CINs (B1 and B2)**	Sum CCINs (C1-C13)**	Sum products (HCLDs, CINs and CCINs)	Products/CLD _{remaining}	Total Mass Recovered
		(μmol)	(μmol)	%	(μmol)	(μmol)	(μmol)	(μmol)	%	%
CONTROLS	G28_poisoned	1.63	1.13	69%	0.002	ND	ND	0.002	0.0%	70%
	G30_poisoned	1.63	0.97	59%	0.001	ND	ND	0.001	0.0%	59%
	G41_poisoned	1.63	1.06	65%	0.004	ND	ND	0.004	0.0%	65%
	G50_poisoned	1.63	0.90	55%	0.002	ND	ND	0.002	0.0%	55%
	G59_poisoned	1.63	1.07	66%	0.001	ND	ND	0.001	0.0%	66%
	AVERAGE	1.63	1.03	63%	0.00	ND	ND	0.002	0%	63%
ORIGINAL MICROCOSMS	G4	3.26	1.36	42%	0.14	0.01	0.11	0.26	19%	50%
	G19	3.26	1.75	54%	0.45	0.04	0.24	0.73	42%	76%
	G31	3.26	1.73	53%	0.13	0.05	0.05	0.22	13%	60%
	G45	3.26	0.83	25%	0.24	0.03	0.05	0.32	39%	35%
	G54	3.26	1.18	36%	0.51	0.04	0.10	0.65	55%	56%
	G63	3.26	1.81	56%	0.03	0.01	0.78	0.82	45%	81%
	G14_TCE	1.62	1.05	65%	0.62	0.09	0.20	0.91	87%	121%
	AVERAGE	3.03	1.39	47%	0.30	0.04	0.22	0.56	43%	68%
MICROCOSM TRANSFERS AMENDED REGULARLY	GT15_TCE	17.9	5.60	31%	2.43	0.65	1.08	4.2	74%	54%
	GT3_TCE	15.8	5.70	36%	4.13	0.12	1.37	5.6	98%	72%
	GT5	23.6	7.37	31%	1.29	0.51	1.12	2.9	40%	44%
	GT20	25.7	8.42	33%	2.71	1.02	1.13	4.9	58%	52%
	GT33	19.6	6.11	31%	1.54	0.23	1.14	2.9	48%	46%
	GT4	22.4	5.99	27%	0.48	0.16	0.63	1.3	21%	32%
	AVERAGE	20.8	6.53	32%	2.10	0.45	1.08	3.6	56%	49%

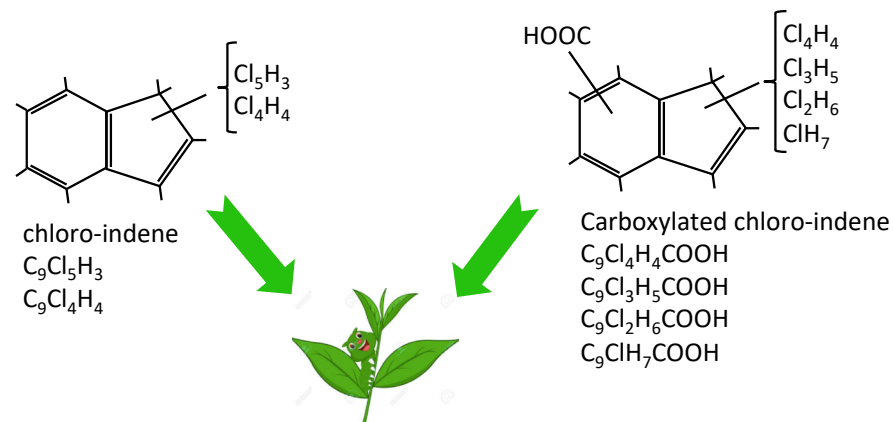
Analysis of field samples

Sample	Chlordecone		Hydrochlordecones	Chloroindenes	
	CLD		MHCLD (A1) & DHCLD (A2)	C ₉ Cl ₅ H ₃ (B1)	C ₉ Cl ₄ H ₄ (B2)
	ng/g solids	area/g solids	%CLD area/g solids	ng/g solids	%B1 area/g solids
Soils:					
River	122	7.2E+06	6%	0.45	ND
Banana	133	1.3E+07	7%	2.1	20%
Stream	1020	9.3E+07	3%	2.1	3.2%
Coco	267	3.1E+07	8%	22	14%
Sea	191	2.9E+07	7%	8.1	23%
Bridge	435	1.8E+07	1%	1.0	ND
Activated Carbon Samples:					
AC-1	24	7.3E+06	10%	ND	ND
AC-2	6032	1.8E+09	8%	ND	ND
AC-3	8416	2.5E+09	4%	ND	ND

Conclusions and Future Work

- Observation of open cage structures in soil samples from Guadeloupe
- Demonstration of almost complete dechlorination in microcosms (up to 9 Chlorine atoms lost)
- Possibility of complete degradation of chlordecone and bioremediation by addition of electron donor substrates – But rates are slow

Need for a better understanding of the mechanism of formation of the indenenes to accelerate the dechlorination process



Acknowledgements



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University des Antilles

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- Region Guadeloupe
- SIAEAG Guadeloupe
- AIP DEMICHLORD

