AECOM

Field-Scale Evaluation of Biosparging to Mitigate Long-Term Dissolution and Mass Discharge of Contaminants from Coal Tar and Creosote

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Risk-Based NAPL Management

Composition

soluble/volatile fractions

- Is there a dissolved-phase risk?
- Is there a vapor-phase risk?

Saturation

- Is NAPL Mobile?
- Is NAPL Migrating?

Management decisions based on a robust NAPL CSM →

Remedial actions that directly and efficiently mitigate risk

Risk-Based NAPL Management

Coal Tar and Creosote Sites

- NAPL is primarily immobile and at residual saturations
- NAPL can be highly weathered
- Primary risk is offsite migration of the dissolved-phase plume
- NAPL is the source of BTEX, PAHs, and/or PCP to groundwater

Risk-Based NAPL Strategy

- Decrease mass discharge to less than the attenuation rate of the dissolved plume

NAPL Remediation Approaches

- Saturation change
- Composition change
- Containment

Risk-Based NAPL Management - Case Studies

Creosote

- Former wood treating facility and mill in Montana
  - Onsite creosote (DNAPL) source area with offsite dissolved plume
  - Primarily pentachlorophenol (PCP) and PAHs (naphthalene)
  - Aerobic biooxidation with biosparging being evaluated (Pilot Study in 2015-2016)

Risk-Based NAPL Management - Case Studies

Coal Tar

- Former Manufactured Gas Plant (MGP) site in Florida
  - Onsite coal tar (DNAPL) source area with offsite dissolved plume
  - VOCs and PAHs
  - Aerobic biooxidation with biosparging at property boundary (testing in portion of source area)

Risk-Based NAPL Management - Case Studies

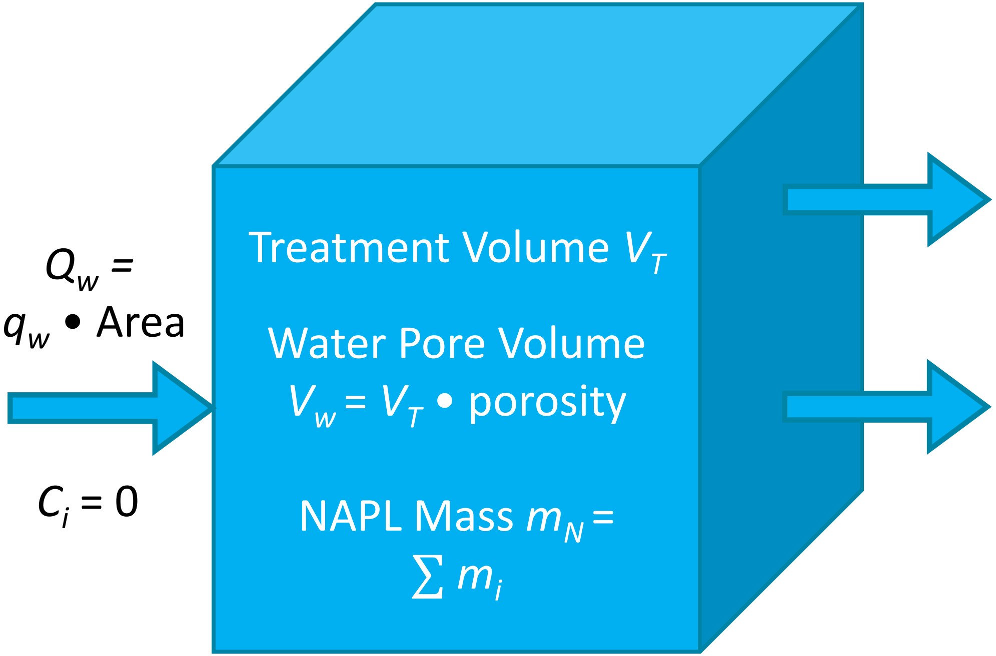
Remediation Objectives

- Change composition of the NAPL by enhancing removal of groundwater contaminants
- Decrease mass discharge to less than the attenuation capacity of the groundwater system
- Contain dissolved plume onsite via natural attenuation

How does biosparging affect NAPL composition?

Can biosparging achieve remediation objectives?

NAPL Depletion Evaluation Approach



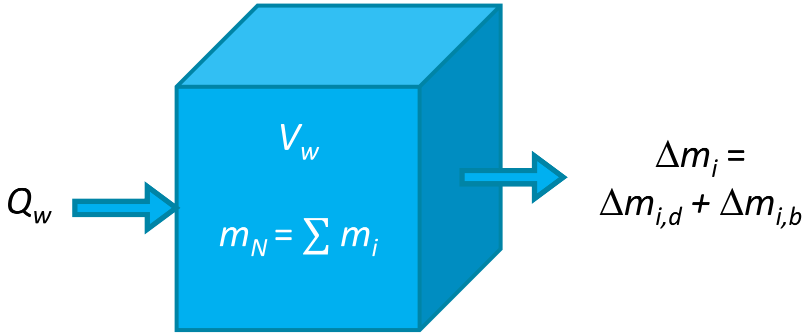
Δm_i = mass loss of compound *i* from the NAPL

Dissolution and Advection
$\Delta m_{i,d} = Q_w \bullet C_i$

Dissolution and Biooxidation
$\Delta m_{i,b} = V_w \bullet C_i \bullet (1 - e^{-k \bullet \Delta t})$

C_i = effective aqueous solubility of compound *i* from the NAPL

NAPL Depletion Evaluation Approach



Inputs

- Treatment volume dimension and hydrologic properties
- NAPL mass and compound mass fractions
- Effective solubility model
- Biooxidation rate of compounds

Excel-Based Numerical Evaluation

- At each time step (Δt)
  - Effective solubility estimated from Raoult's Law and current NAPL composition
  - NAPL composition changes as compounds are removed

Approach Assumes

- Equilibrium dissolution
- Homogeneity

Solubility Modeling

Raoult's Law

The effective aqueous solubility of compound *i* from the NAPL is

$$C_i = C_s^i \frac{\chi_i}{FR_i}$$

C_s^i = pure phase aqueous solubility of compound *i*

χ_i = mole fraction of compound *i* in NAPL

FR_i = solid-liquid fugacity ratio of compound *i*

Mole Fraction

$$\chi_i = C_N^i \frac{\boldsymbol{MW_N}}{MW_i}$$

C_N^i = mass fraction of compound *i in NAPL*

MW_i = molecular weight of compound *i*

MW_N = average molecular weight of the NAPL

Solubility Modeling

Raoult's Law-Based Method for Determination of Coal Tar Average Molecular Weight

Brown et al. 2005. *Environmental Toxicology and Chemistry*, Vol. 24, No. 8, pp. 1886-1892

Laboratory Method

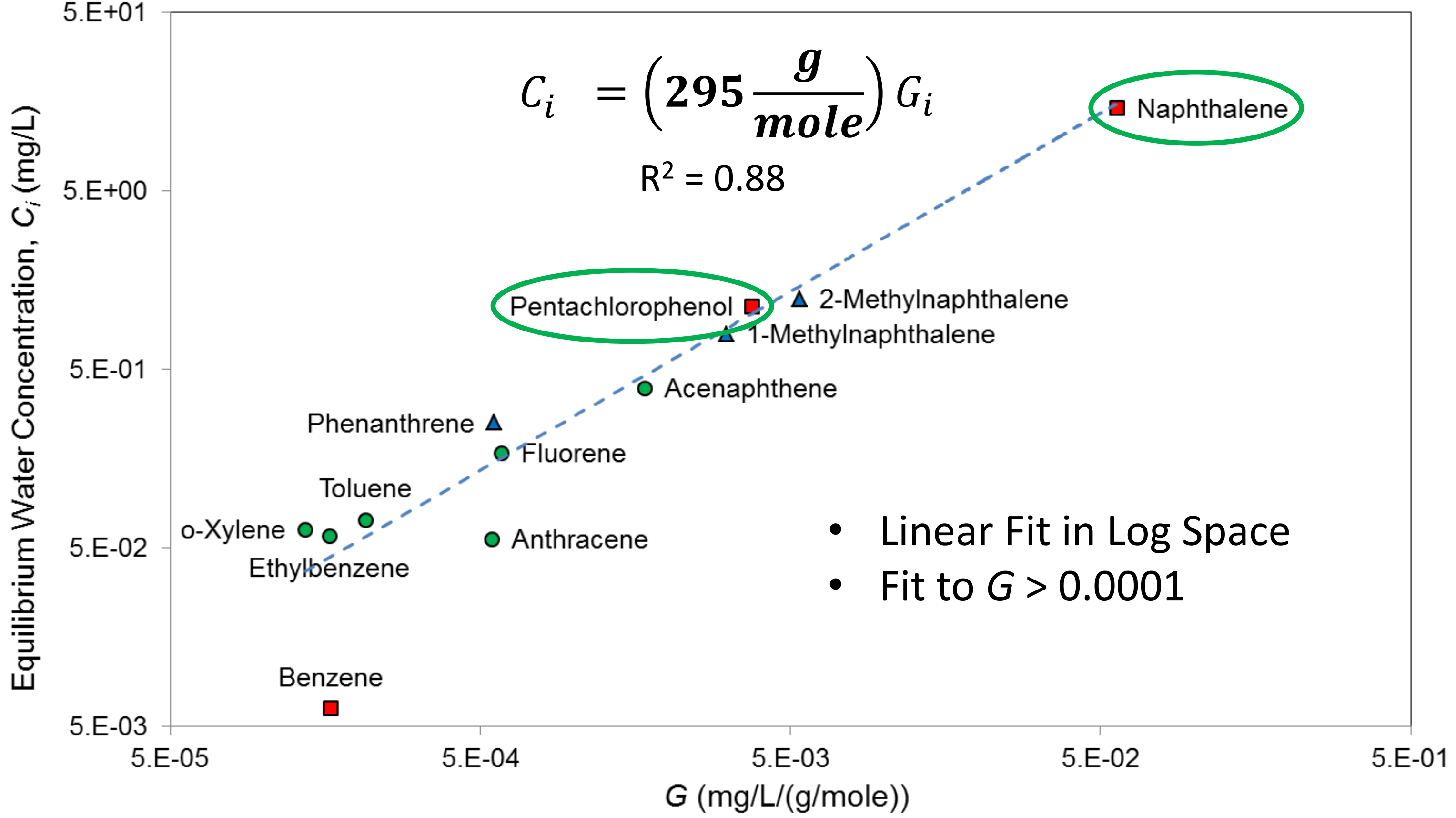
- Mass fraction of target compounds in the NAPL
- NAPL-water equilibrium studies to quantify effective aqueous solubility of target compounds

Key Concept: Linear slope of effective solubility for target compounds versus rearrangement of Raoult's Law is the average molecular weight of the NAPL

$$C_i = \boldsymbol{MW_N} G_i$$

$$G_i = \frac{C_s^i}{FR_i}\frac{C_N^i}{MW_i}$$

Solubility Modeling – Creosote Case Study

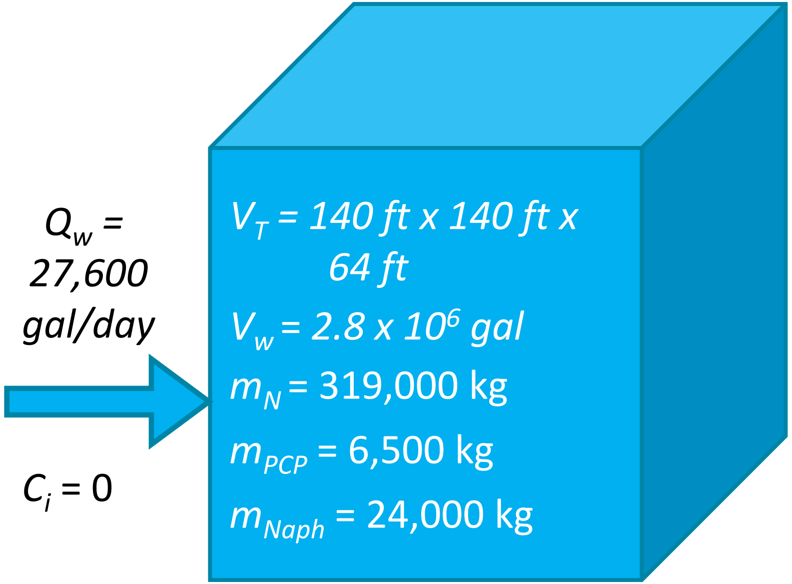


- Linear Fit in Log Space
- Fit to $G > 0.0001$

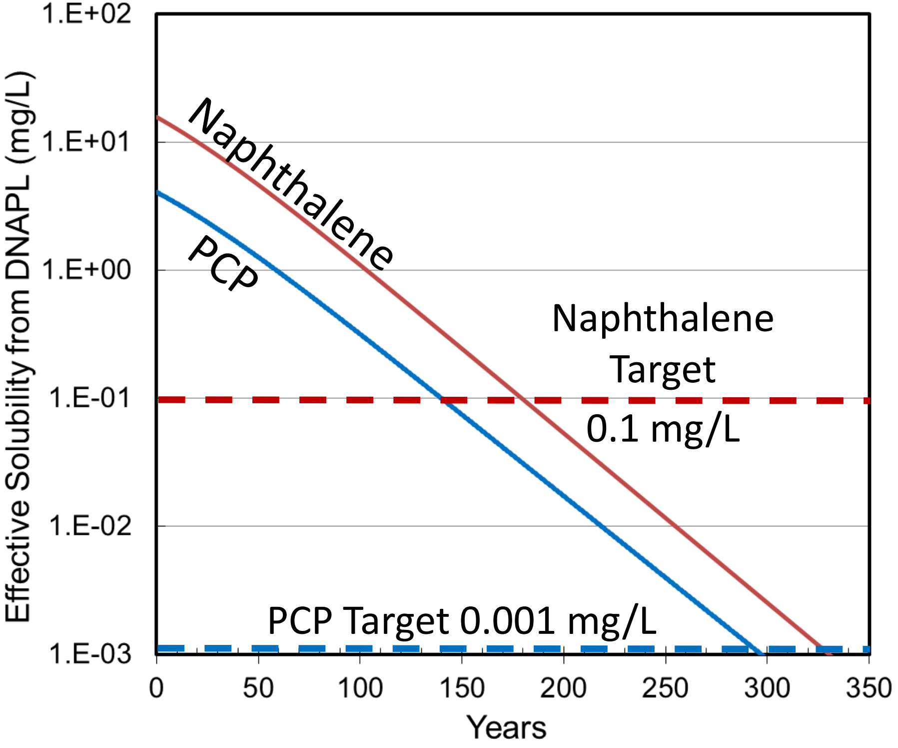
NAPL Depletion Evaluation – Creosote Case Study

Solubility Model:

$$C_i = C_N^i \frac{C_s^i}{FR_i} \frac{295 \frac{\text{g}}{\text{mole}}}{MW_i}$$



Dissolution and Advection $\Delta m_{i,d} = Q_w \bullet C_i$



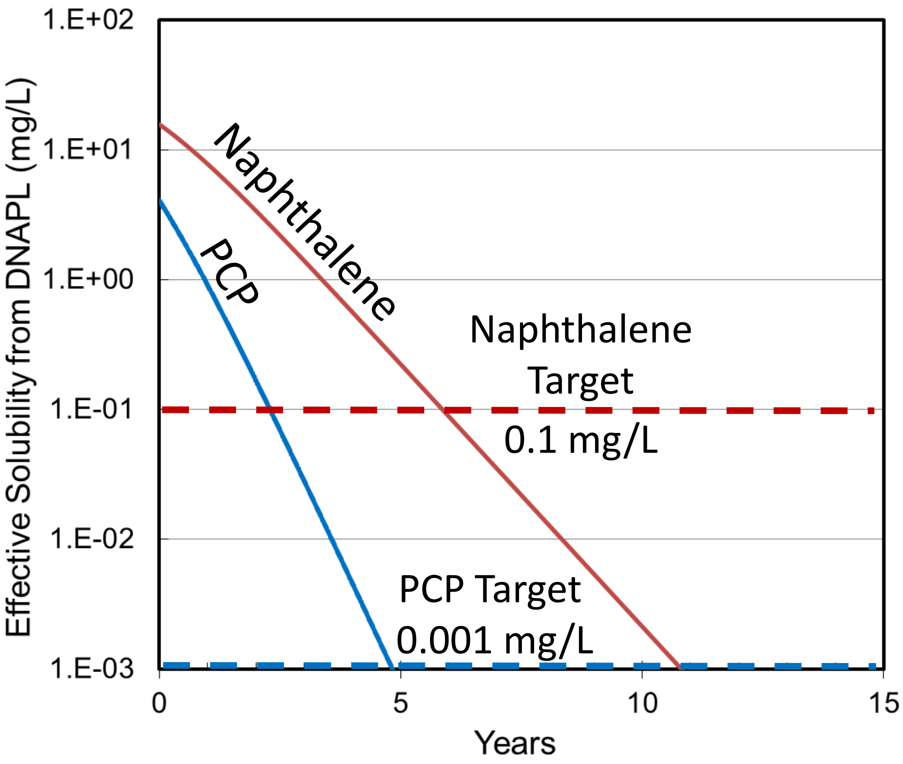
NAPL Depletion Evaluation – Creosote Case Study

Dissolution and Biooxidation

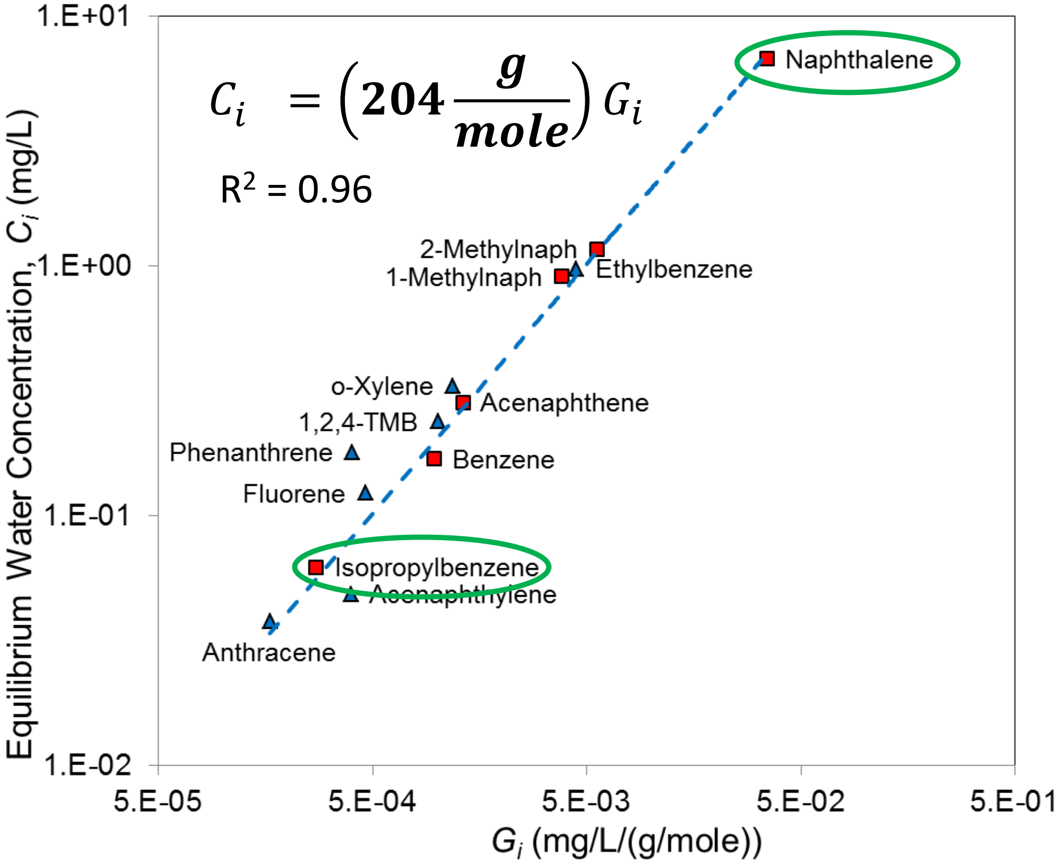
$$\Delta m_{i,b} = V_w \bullet C_i \bullet (1 - e^{-k \bullet \Delta t})$$

Field-Scale Biosparging Study

- Fit rates (k) to mass fraction reduction from soil data (baseline vs. 270-day)
- Naphthalene
  - 39% decrease
  - Half-life = 2 days, k = 0.35/d
- PCP
  - 66% decrease
  - Half-life = 0.7 days, k = 0.95/d
- Onsite aerobic bioreactor, half-life = 0.1 day

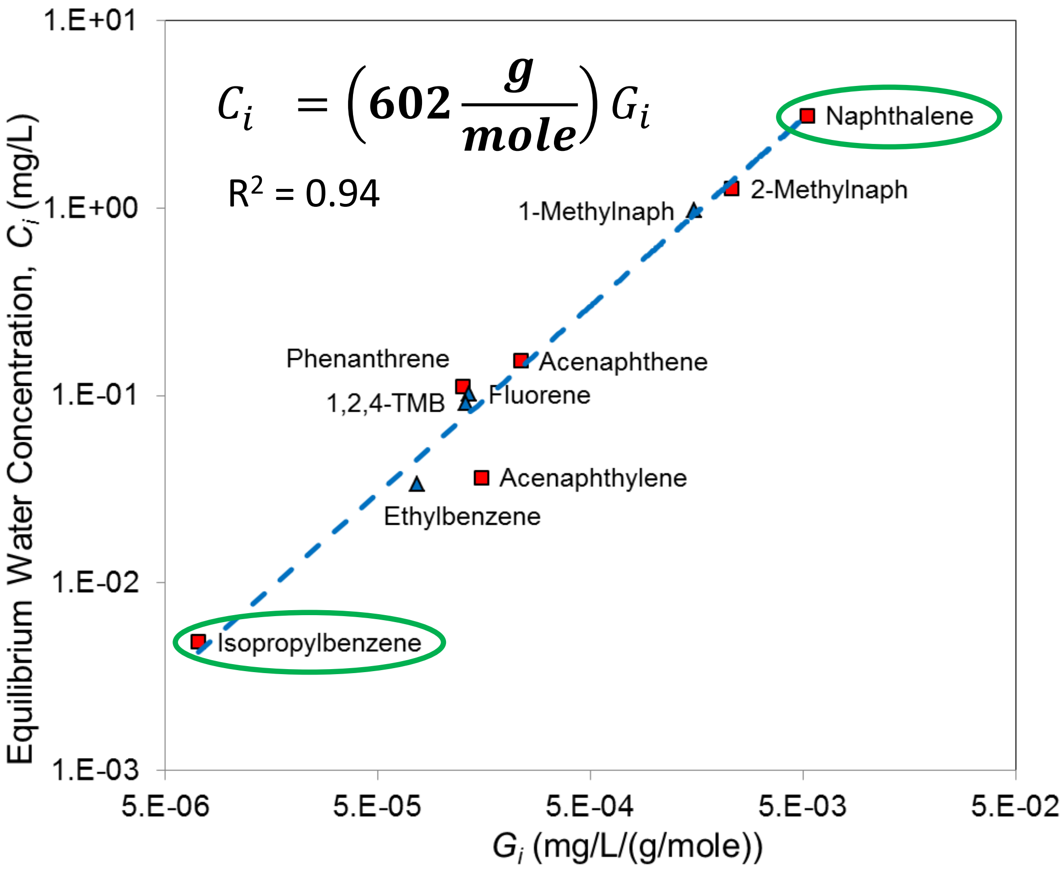


Solubility Modeling – Coal Tar Case Study



- DNAPL outside of biosparge treatment area
- Mass Fractions
  - 4.4% Naphthalene
  - 0.05% Isopropylbenzene
  - 66% TPH (C8-C40)

Solubility Modeling – Coal Tar Case Study

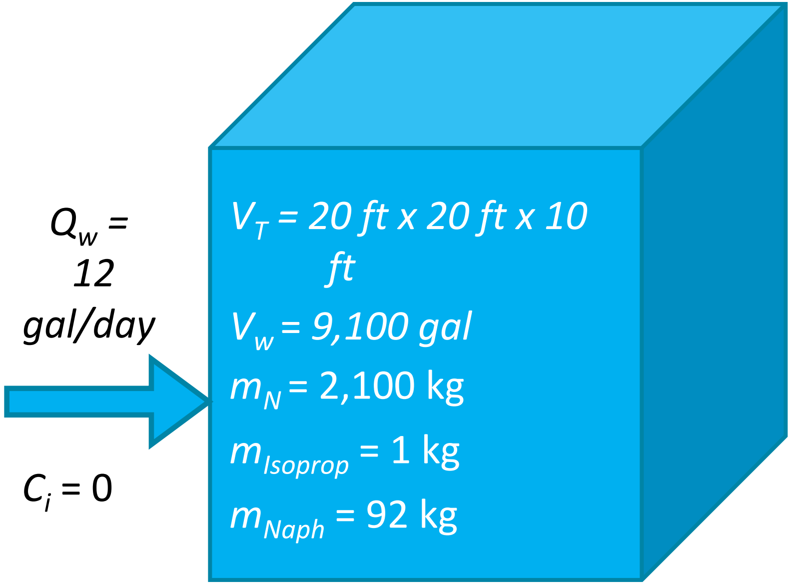


- DNAPL within biosparge treatment area for 1 year
- Mass fractions
  - 0.7% Naphthalene
  - 0.001% Isopropylbenzene
  - 22% TPH (C8-C40)
- Decrease in mass fraction
  - 85% Naphthalene
  - 97% Isopropylbenzene
  - 67% TPH (C8-C40)

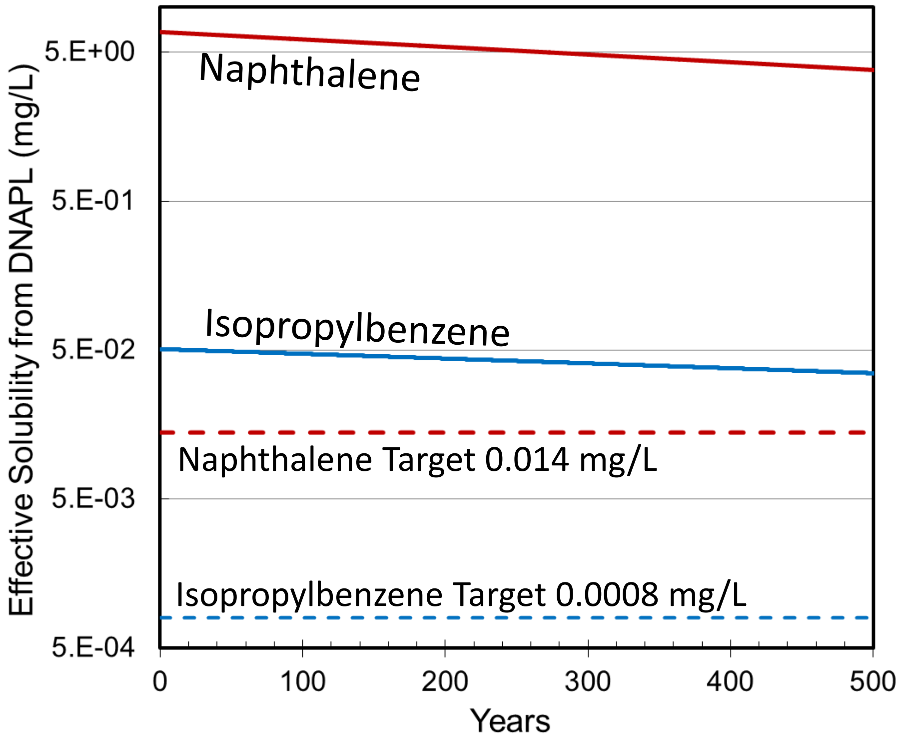
NAPL Depletion Evaluation – Coal Tar Case Study

Solubility Model:

$$C_i = C_N^i \frac{C_s^i}{FR_i} \frac{204\,\frac{\text{g}}{\text{mole}}}{MW_i}$$



Dissolution and Advection $\Delta m_{i,d} = Q_w \bullet C_i$



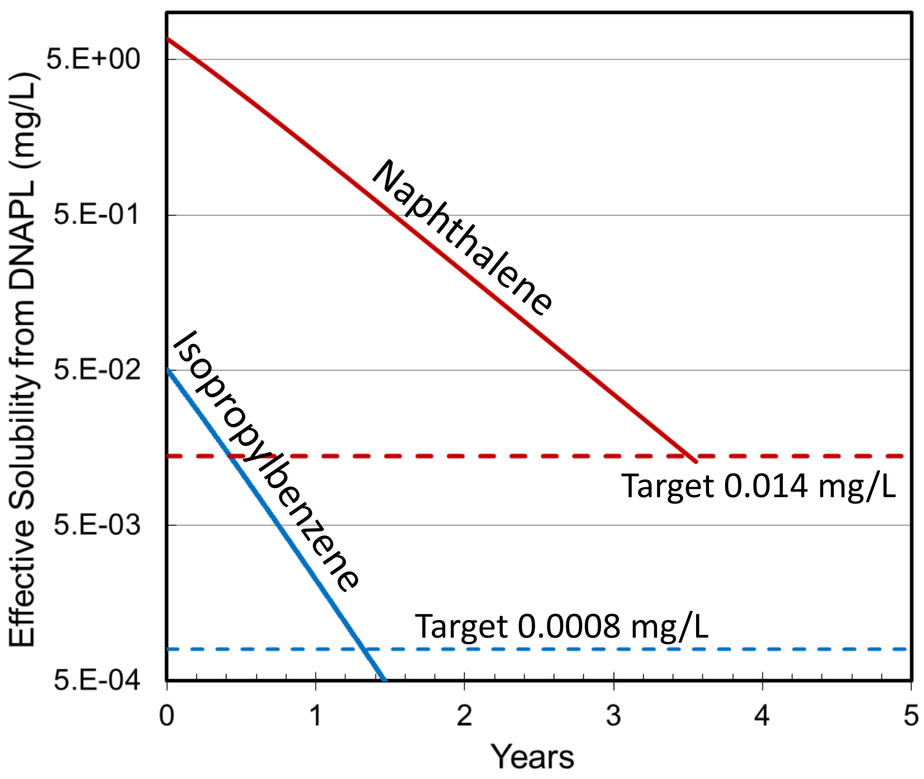
NAPL Depletion Evaluation – Coal Tar Case Study

Dissolution and Biooxidation

$$\Delta m_{i,b} = V_w \bullet C_i \bullet (1 - e^{-k \bullet \Delta t})$$

Field-Scale Biosparging Study

- Fit rates (k) to mass fraction reduction in NAPL samples (1 year)
- Naphthalene
  - 85% decrease
  - Half-life = 0.35 days, k = 2/d
- Isopropylbenzene
  - 97% decrease
  - Half-life = 0.1 days, k = 6.9/d
- Partitioning to air ignored



Summary

- A laboratory-based Raoult's Law solubility model provides a basis for modeling long-term NAPL dissolution
- Biooxidation processes enhance NAPL dissolution and weathering
- Simple mass-balance models are viable tools to evaluate remedial alternatives in the context of an FFS

Take Home: Dissolved-phase remediation strategies (including chemical and biological oxidation) are viable alternatives to enhance NAPL composition change and mitigating long-term dissolution from NAPL

- Cost effectively
- Reasonable time

Thank You!

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