



# Improving Dissolved Organic Chemical Concentration Measurements at Groundwater/Surface-Water Interfaces Containing NAPL



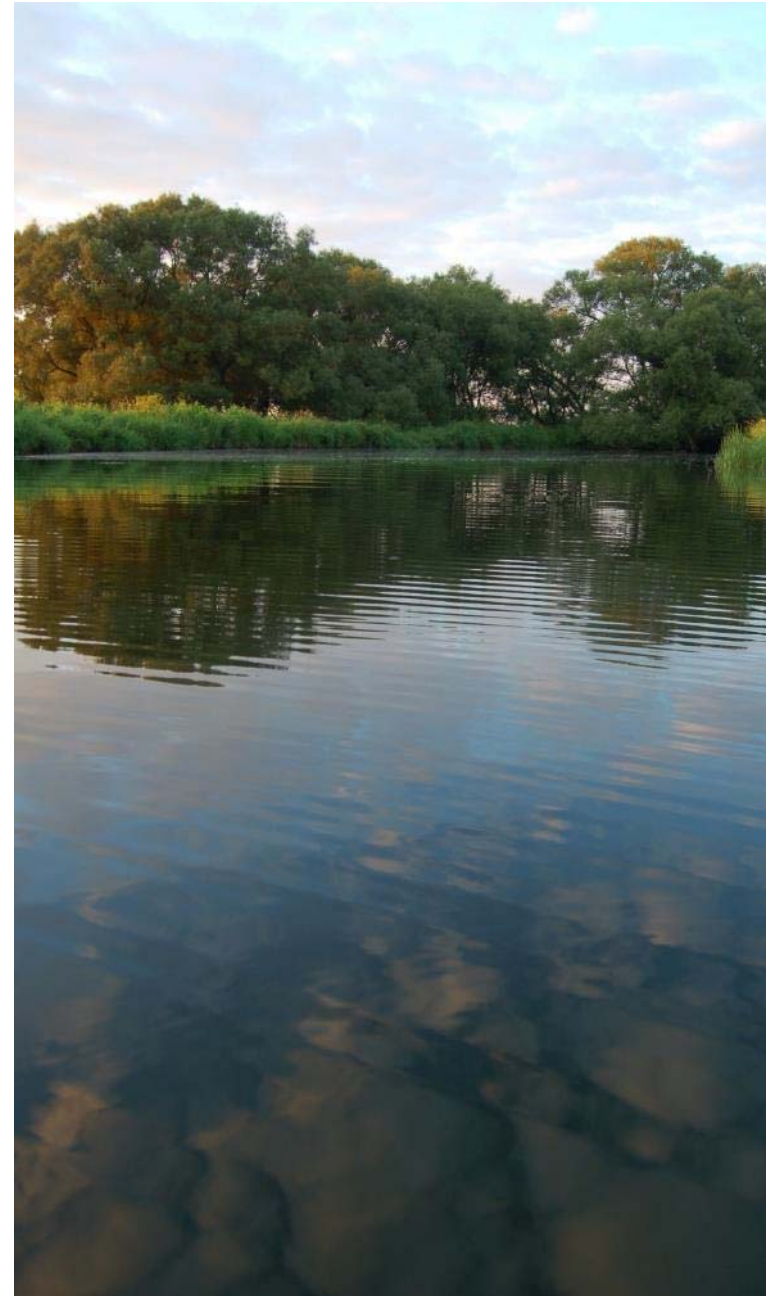
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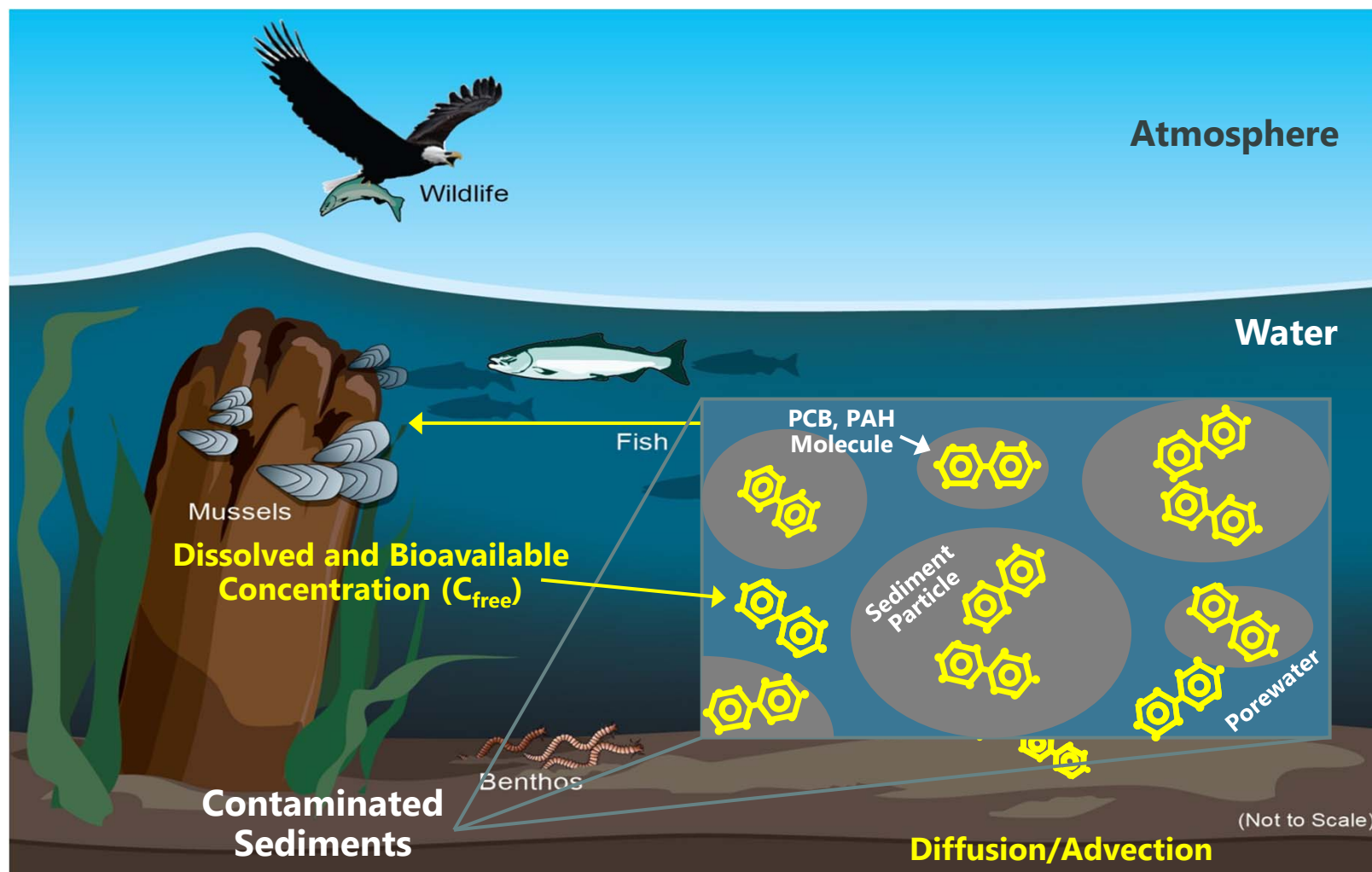
Battelle Eleventh International Conference on Remediation of Chlorinated and Recalcitrant Compounds  
April 8–12, 2018 — Palm Springs, California, USA

# Outline

- Importance of accurate aqueous-phase samples
- Complexities due to nonaqueous phase liquid (NAPL)
- NAPL exclusion concepts
- Chemical sampling tests
- Possible applications
- Summary and conclusions



# Importance of Accurate Aqueous Samples

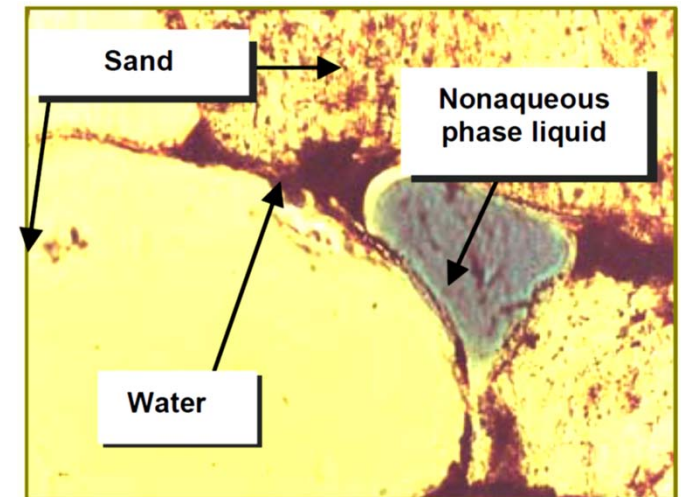


Source: Burgess, R.M., 2013. *Passive Sampling for Measuring Freely Dissolved Contaminants in Sediments: Concepts and Principles*. Training Slides from 23rd Annual NAPRM Training. U.S. Environmental Protection Agency ORD NHEERL. Available at: [https://clu-in.org/conf/tio/Porewater2\\_111914/resource.cfm](https://clu-in.org/conf/tio/Porewater2_111914/resource.cfm).



# NAPL Can Exaggerate “Aqueous” Concentrations

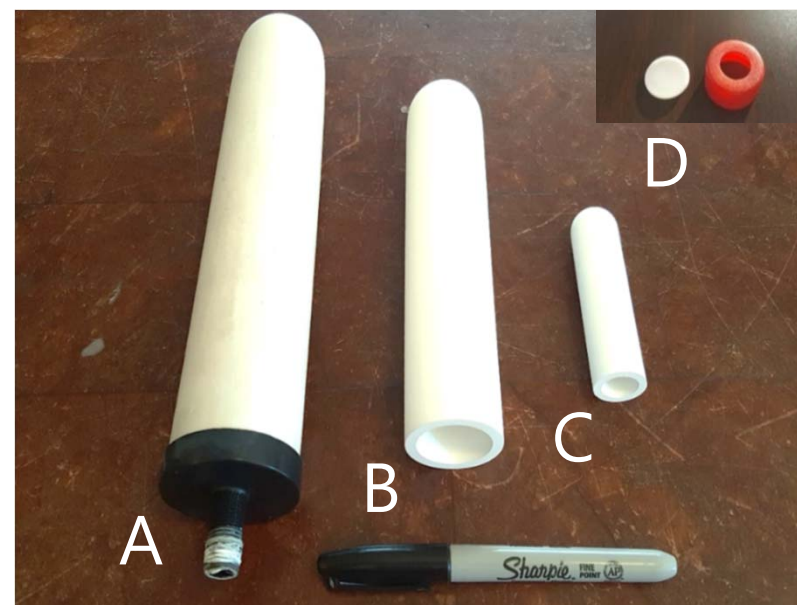
- NAPL enters pore-fluid samplers.
- NAPL coats hydrophobic passive samplers.
- Aqueous concentrations calculated from sediment samples can exceed effective solubility.
- **Presence of NAPL can result in porewater concentrations that are biased high—above true dissolved, bioavailable concentrations.**



Bottom figure from: Wilson, J.L., S.H. Conrad, W.R. Mason, W. Peplinski, and E. Hagan, 1990. *Laboratory Investigation of Residual Liquid Organics from Spills, Leaks, and the Disposal of Hazardous Wastes in Groundwater*. EPA/600/6-90/004. April 1990.

# Porous, Hydrophilic Capillary Barriers

- **Ceramics**
- Bentonite
- Silica Flour
- Others?



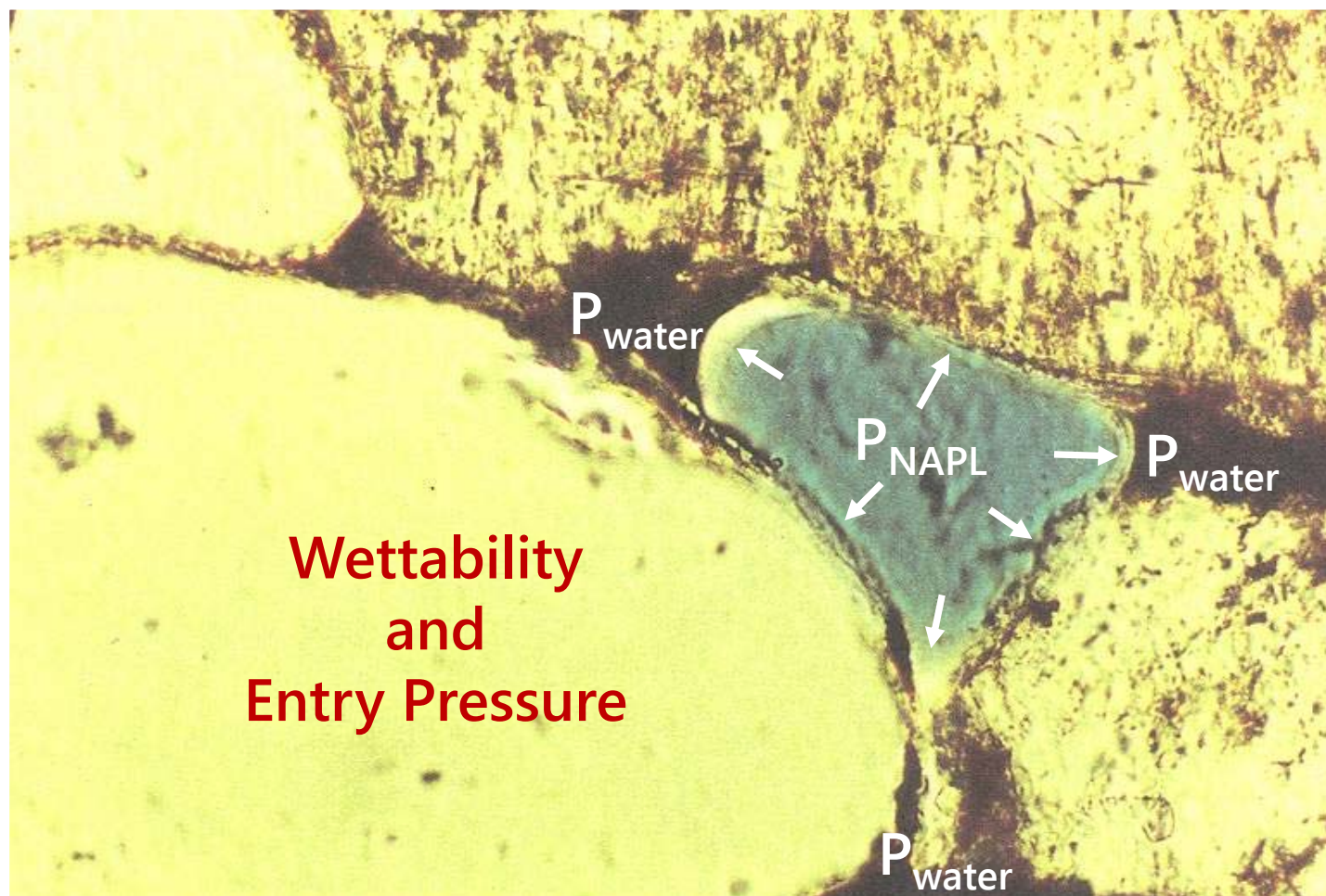
ID	Shape	Pore Size (μm)	K (cm/s)	Porosity	Length (cm)	Outer Diameter (cm)	Approximate Cost (US \$)
A*	Tube	11.2	$8 \times 10^{-5}$	0.22	24	4.9	\$20
B	Tube	2.5	$9 \times 10^{-6}$	0.45	17	4.0	\$100
C	Tube	2.5	$9 \times 10^{-6}$	0.45	8.9	2.2	\$40
D	Disk	2.5	$9 \times 10^{-6}$	0.45	NA	2.2	\$40

Notes:

\* = Physical parameters estimated based on laboratory testing by Anchor QEA. All others provided by manufacturer.

K = hydraulic conductivity

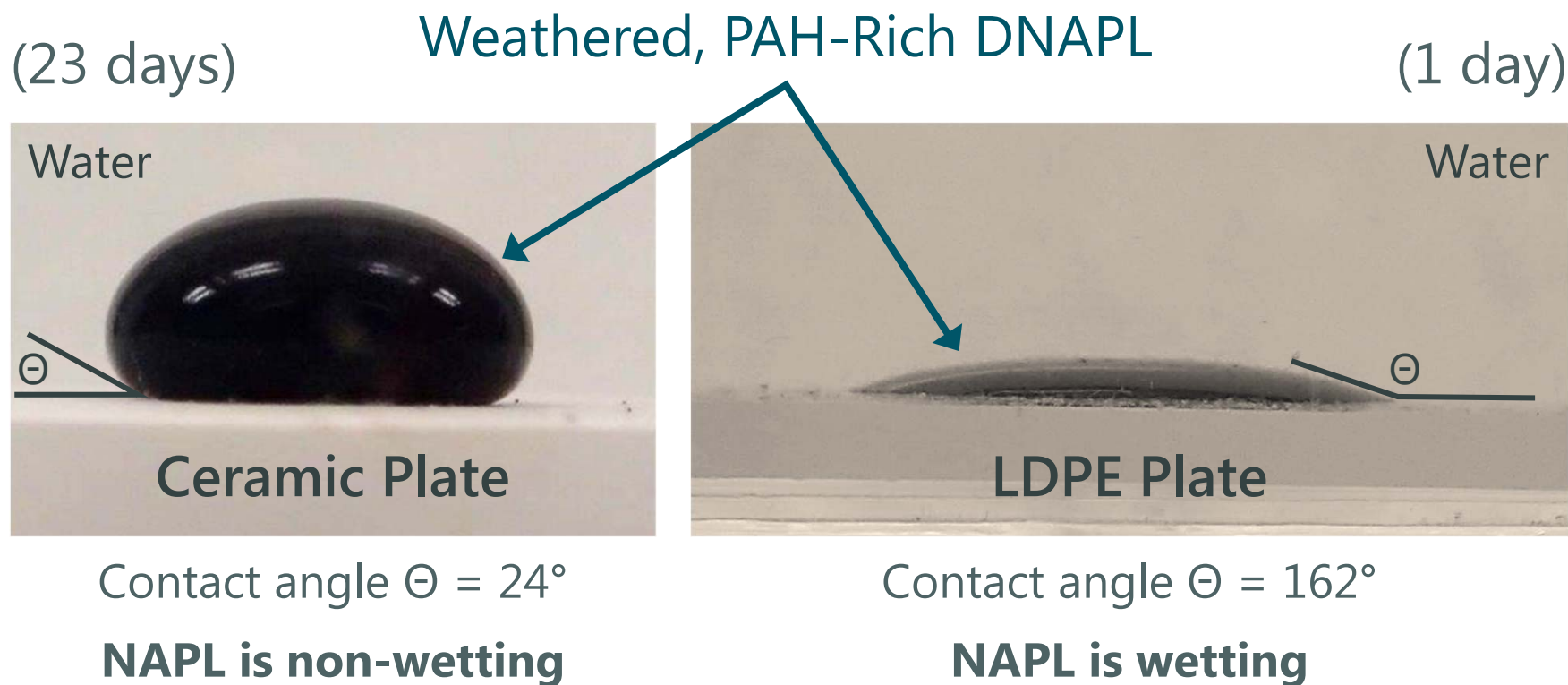
# Fundamentals of NAPL Exclusion



Source: Wilson, J.L., S.H. Conrad, W.R. Mason, W. Peplinski, and E. Hagan, 1990. *Laboratory Investigation of Residual Liquid Organics from Spills, Leaks, and the Disposal of Hazardous Wastes in Groundwater*. EPA/600/6-90/004. April 1990.

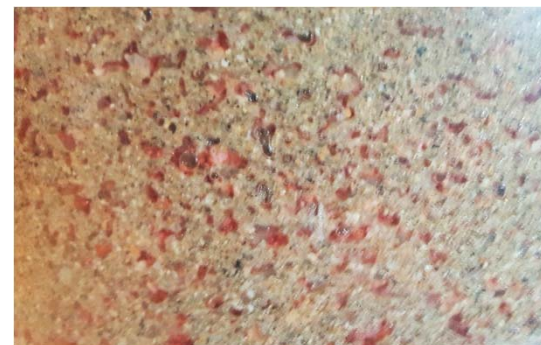
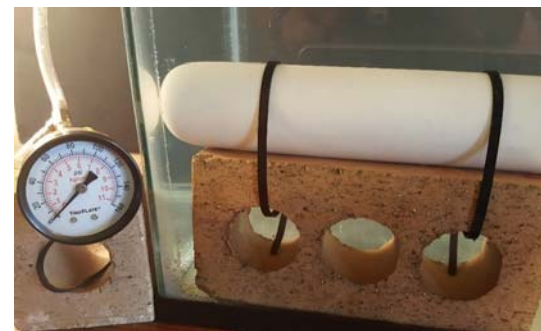


# Comparative Wettability Tests—Dense NAPL on Ceramic and Low Density Polyethylene (LDPE)



# Entry Pressure and NAPL Exclusion Tests

- Measured entry pressure using air pressure bubbling tests
- Tested water pumping in well-graded, fine-to-course sand and 25% to 50% NAPL saturation ( $S_n$ )
  - $S_n = 0.25$ : pumped **25 mL/min** water flow with no sheen or NAPL in effluent—**potentially useful**
  - $S_n = 0.50$ : Sheen in effluent with only 1.5 mL/min water flow—**impractical**





# Depth Below Top of DNAPL Pool Required for Coal Tar/Creosote to Enter Ceramic Pores Without Water Pumping

$$Z_n = (2\sigma \cos \varphi) / [r g (\rho_n - \rho_w)]$$

$Z_n$  = critical DNAPL height above ceramic sampler (cm)

$\sigma$  = NAPL-water interfacial tension (20 dynes/cm = 20 g/s<sup>2</sup>)

$\varphi$  = contact angle (24°)

$r$  = pore radius (1.25 to 5.6 microns = 0.000125 to 0.00056 cm)

$g$  = gravitational constant (980 cm/s<sup>2</sup>)

$\rho_n$  = non-wetting phase (NAPL) density (1.07 g/cm<sup>3</sup>)

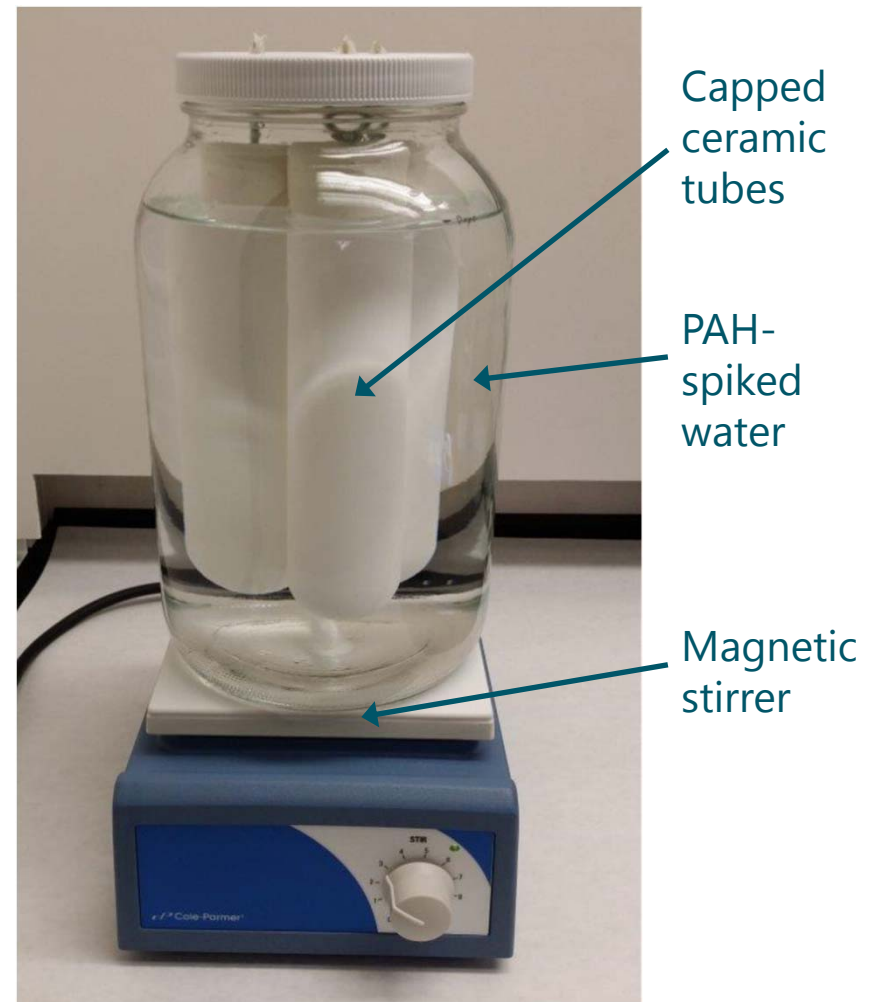
$\rho_w$  = wetting phase (water) density (1.0 g/cm<sup>3</sup>)

$$Z_n = 10 \text{ to } 40 \text{ meters}$$

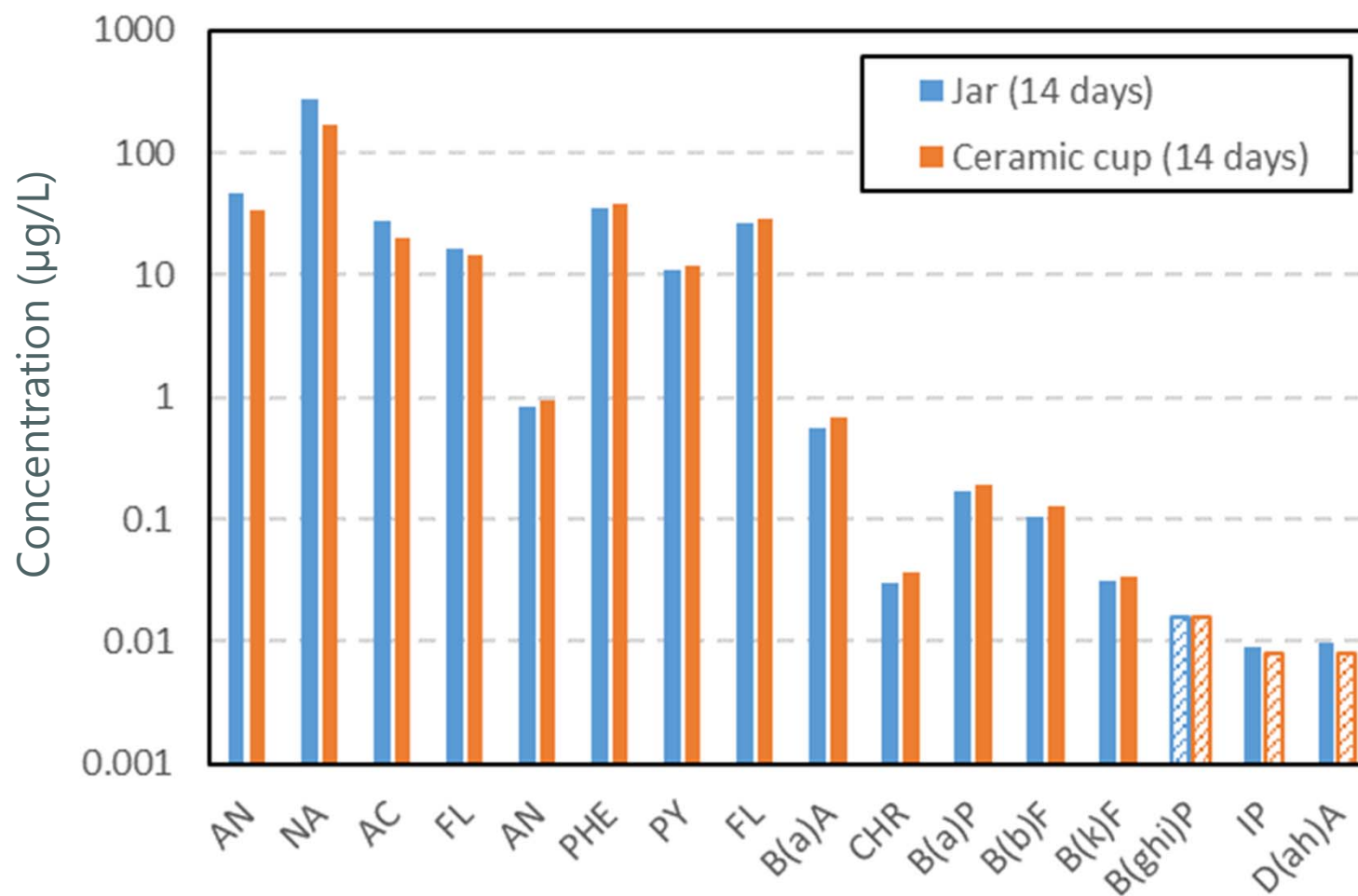
Source: Cohen, R.M., and J.W. Mercer, 1993. *DNAPL Site Evaluation*. C.K. Smoley, Boca Raton, Florida.

# PAH Equilibration Test (No NAPL)

- 16 priority PAHs spiked in water in a 2-L jar
- Porous ceramic cups each containing 120 mL deionized water submerged in jar
- Water in the jar was slowly stirred by a magnetic stir bar and stored in the dark at 20 °C
- Diffusion-based equilibration



# PAH Equilibration, 14-Day Results (No NAPL)



Note: Striped pattern bars indicate method detection level.

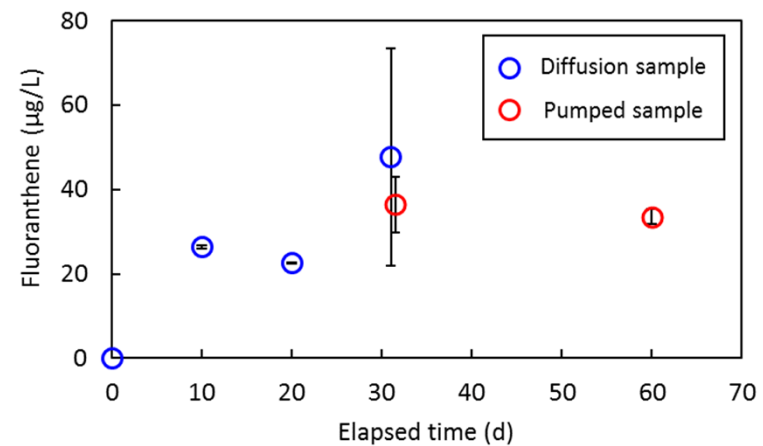
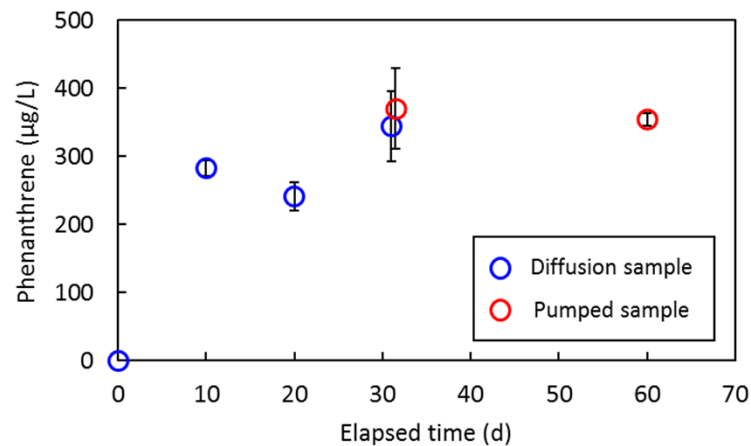
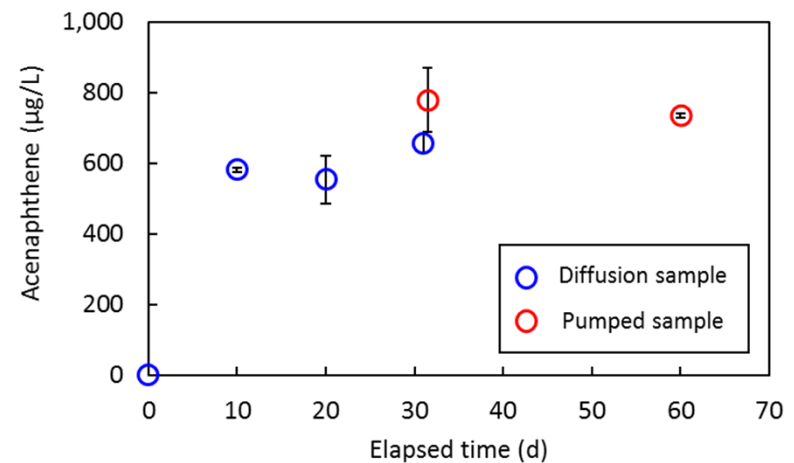
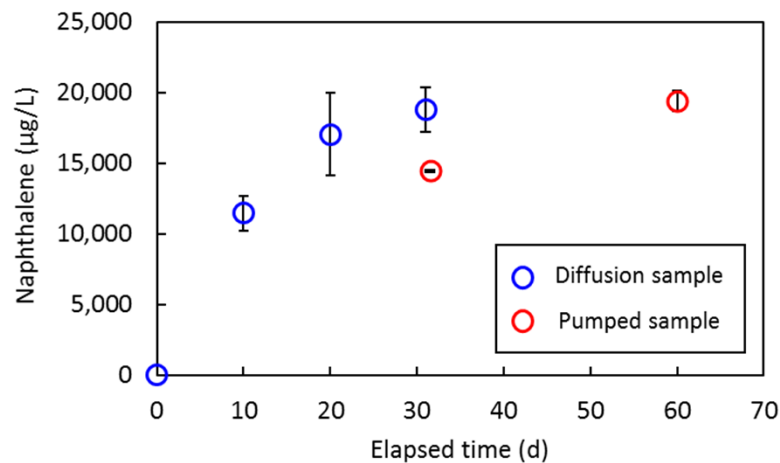


# Porewater Sampling Tests With Diffusive Equilibration and Pumping (With NAPL)

- Aquarium with well-graded sand, 0.5M NaCl water, and 9% creosote NAPL saturation
- Duplicate samples:
  - NAPL-coated sand at 0 and 31 days
  - Diffusion-based water samples at 10, 20, and 31 days
  - Pumped water samples also collected from ceramic tubes at 31 days and 60 days

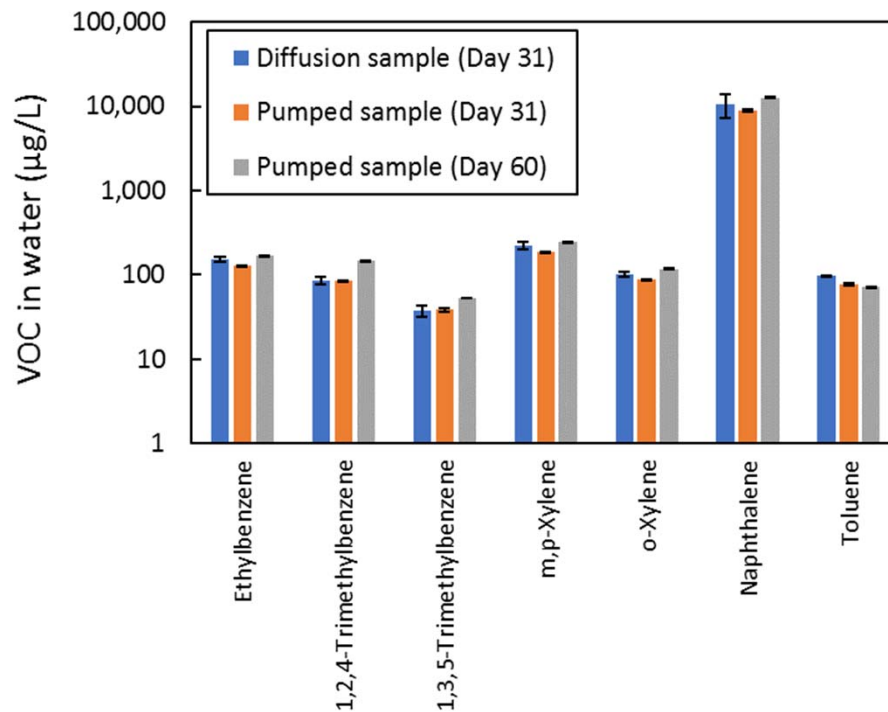


# Porewater Sampling Tests With Diffusive Equilibration and Pumping (With NAPL)

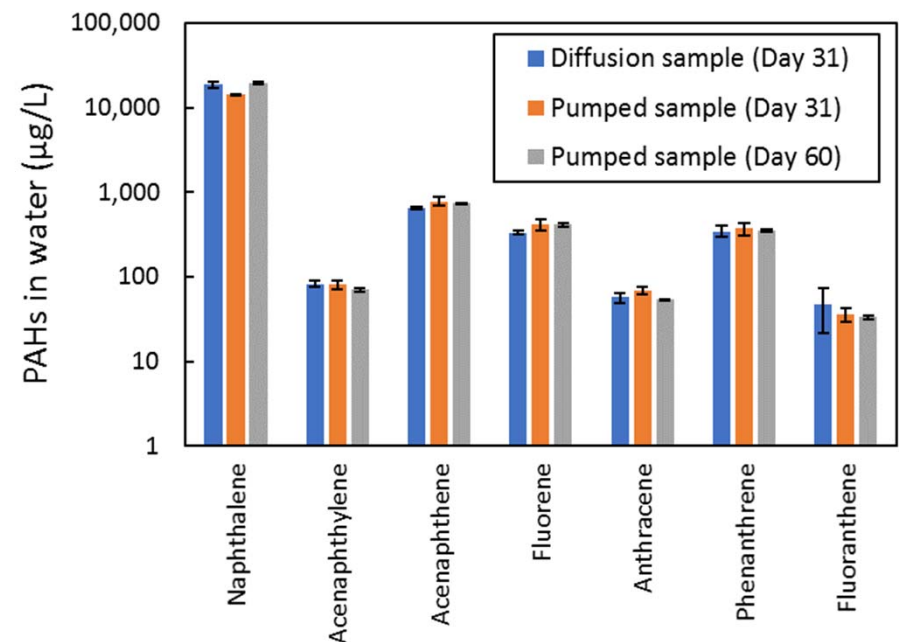


# Porewater Sampling Tests With Diffusive Equilibration and Pumping (With NAPL)

## VOCs



## PAHs





# Potential Uses of Capillary Barrier Materials for Water Sampling Without NAPL Impacts

- Sample porewater by diffusion-based equilibration.
- Protect hydrophobic, sorption-based samplers.
- Pump water samples through capillary barrier in situ (push-point sampler) or ex situ (water filter) to exclude NAPL.
- Use capillary barrier devices in wells with NAPL.

# Summary and Conclusions

- Aqueous concentrations drive risk and remediation.
- Any NAPL in samples can severely bias interpreted aqueous concentrations.
- Capillary barrier materials can be used to sample aqueous phase and avoid impacts due to NAPL, even when directly contacting NAPL.
- Wettability and entry pressure of porous ceramics appear favorable—also readily available and economical.
- Sampling by PAH diffusive equilibration and pumping through ceramic has been demonstrated.

# Next Step

- Field application!



# Acknowledgements

- Anchor QEA Innovation Program



# Questions/Discussion

