

# Ensuring CO<sub>2</sub> Storage and Groundwater Protection Through Water Monitoring: Applying Geochemical and Statistical Tools



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The background of the slide is a photograph of several high-voltage electrical transmission towers (pylons) silhouetted against a sunset sky. The sky is filled with soft, colorful clouds in shades of orange, pink, and blue. The power lines stretch across the frame, connecting the towers.

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# Carbon Sequestration

## Importance of Monitoring

- Ensures CO<sub>2</sub> storage
- Protects valuable assets
- Provides assurance to community

## Geochemical Monitoring for MVA

- Utilizes existing infrastructure
- Employs developed tools
- Provides source attribution

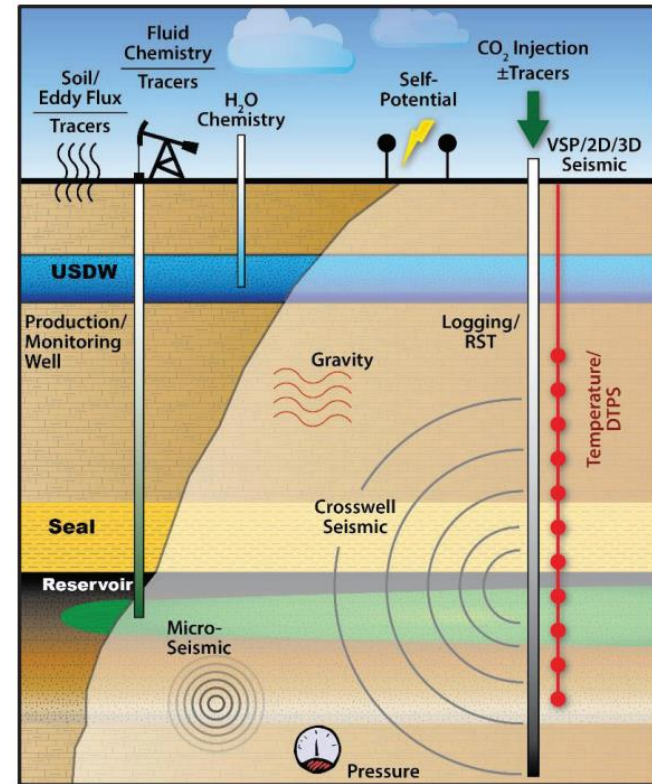
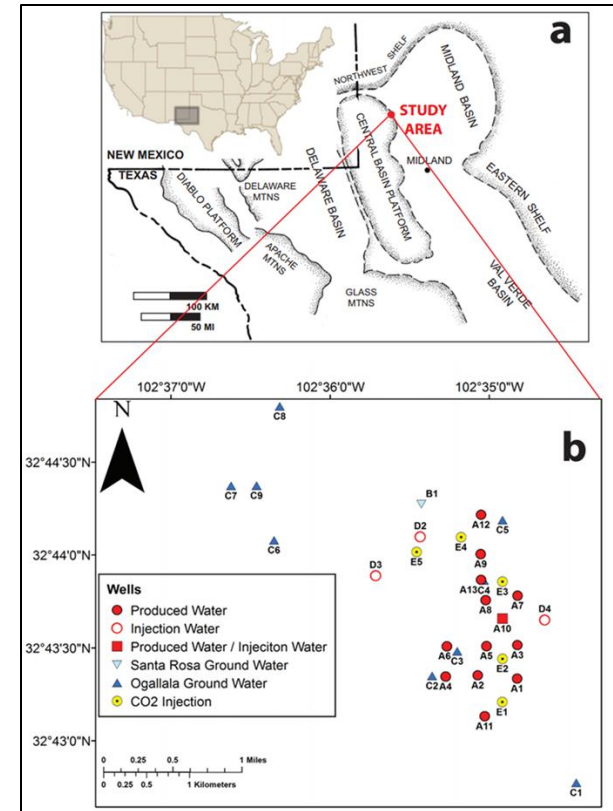
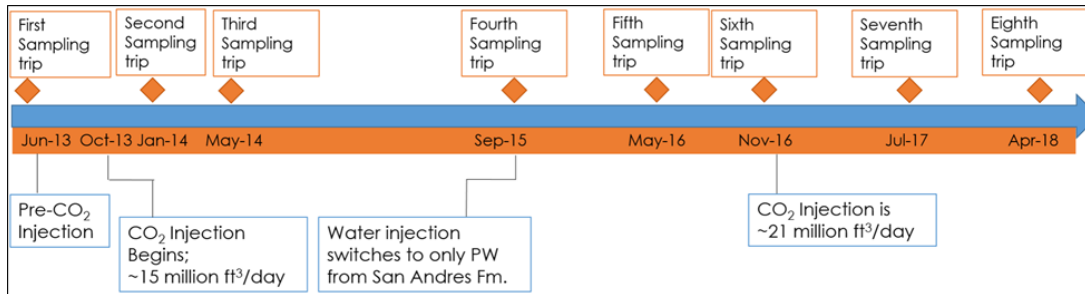


Figure from Balch et al., 2017

# Field Area

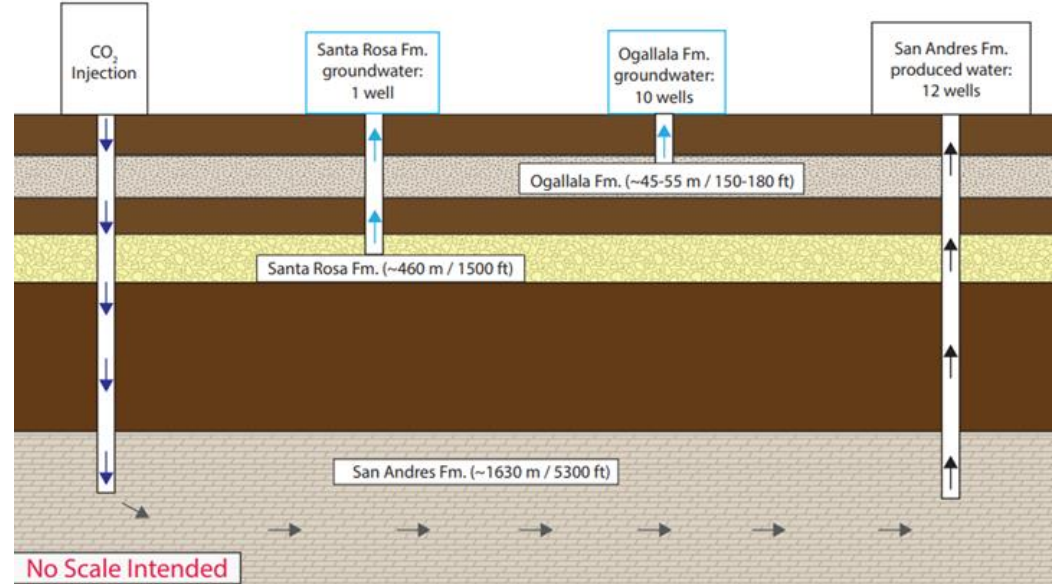
- CO<sub>2</sub> EOR as an analog
- Sampled oil field in Permian Basin, West Texas
  - June 2013 to April 2018
    - Eight field sampling events
  - CO<sub>2</sub> EOR: October 2013
    - One pre-CO<sub>2</sub> injection event



# Field Area

## Samples

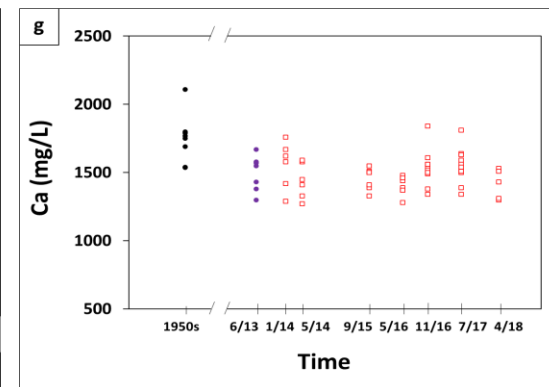
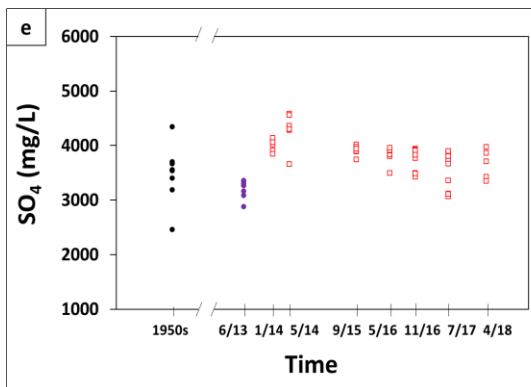
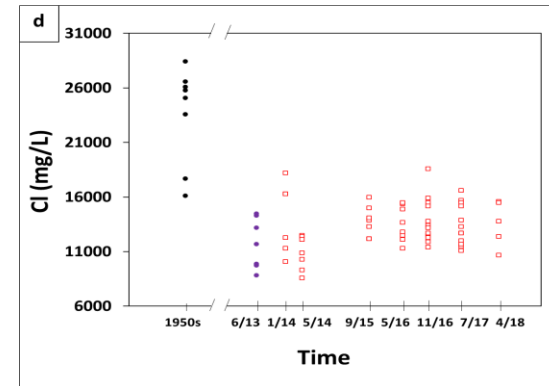
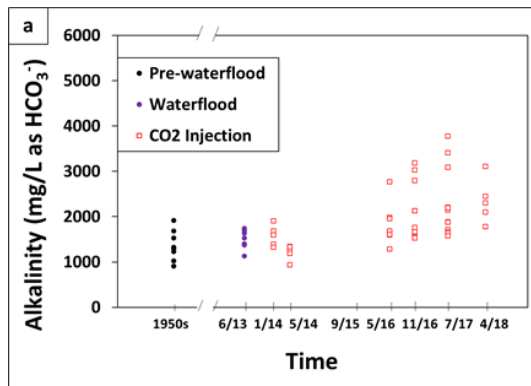
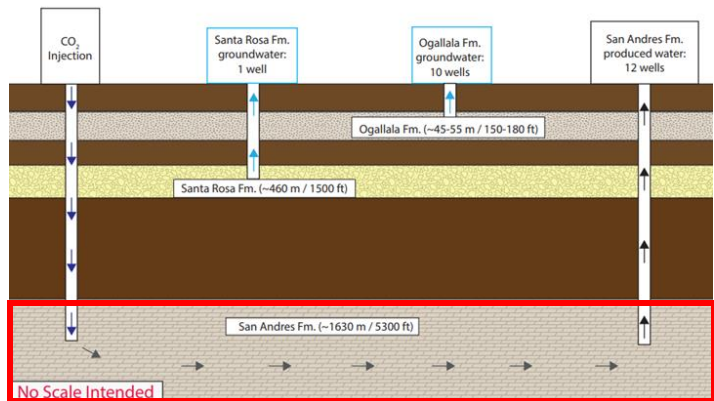
- Ogallala Fm. groundwater
- Santa Rosa Fm. groundwater
- San Andres Fm. produced water



- Developed analytical process for fossil fuel wastewaters
  - Miller et al., 2022

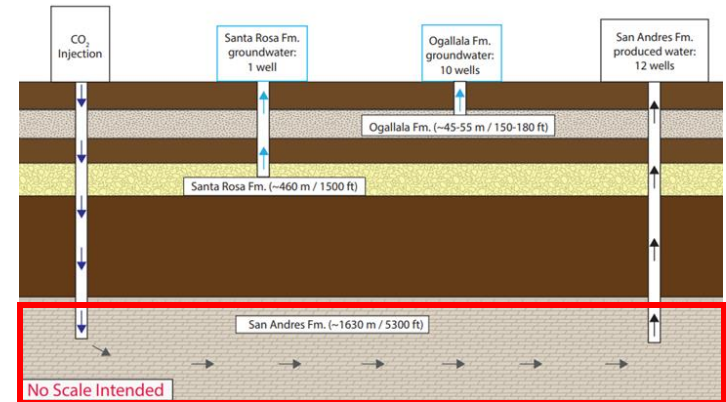
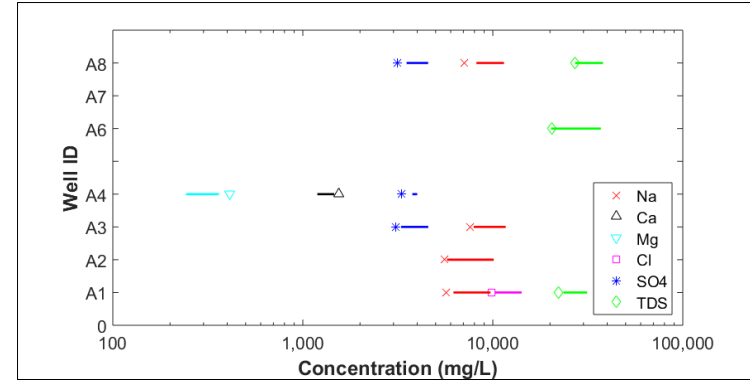
# Results: Producing Formation

- Focus on alkalinity and dissolved inorganics
- Following CO<sub>2</sub> Injection
  - Increases in certain parameters
  - Statistical significance



# Results: Producing Formation

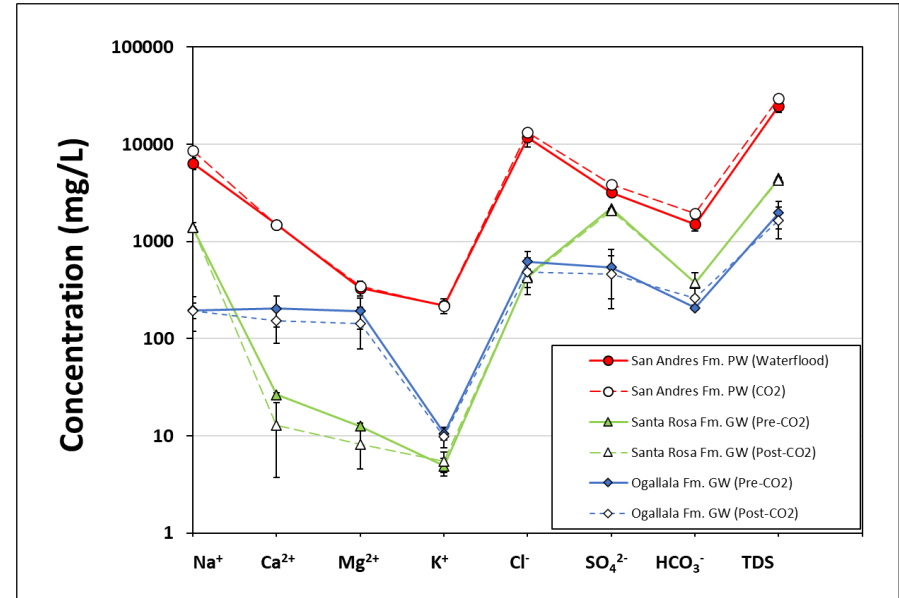
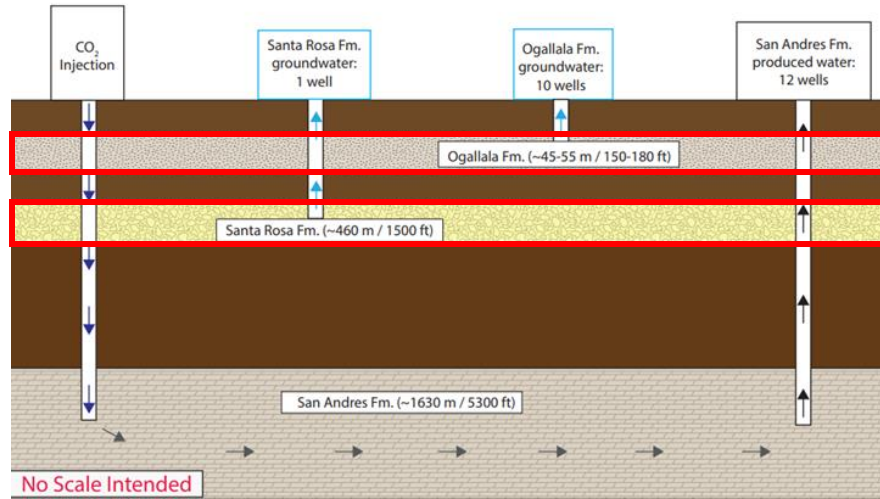
- **Statistical analysis**
  - Generated prediction intervals
  - Significant change: alkalinity, TDS, Na<sup>+</sup>, Cl<sup>-</sup>, and SO<sub>4</sub><sup>2-</sup>
- **Interpretation**
  - CO<sub>2</sub> injection caused significant changes in produced water parameters
  - However, CO<sub>2</sub> injection did not result in significant carbonate dissolution
  - Therefore, reservoir integrity was preserved during injection





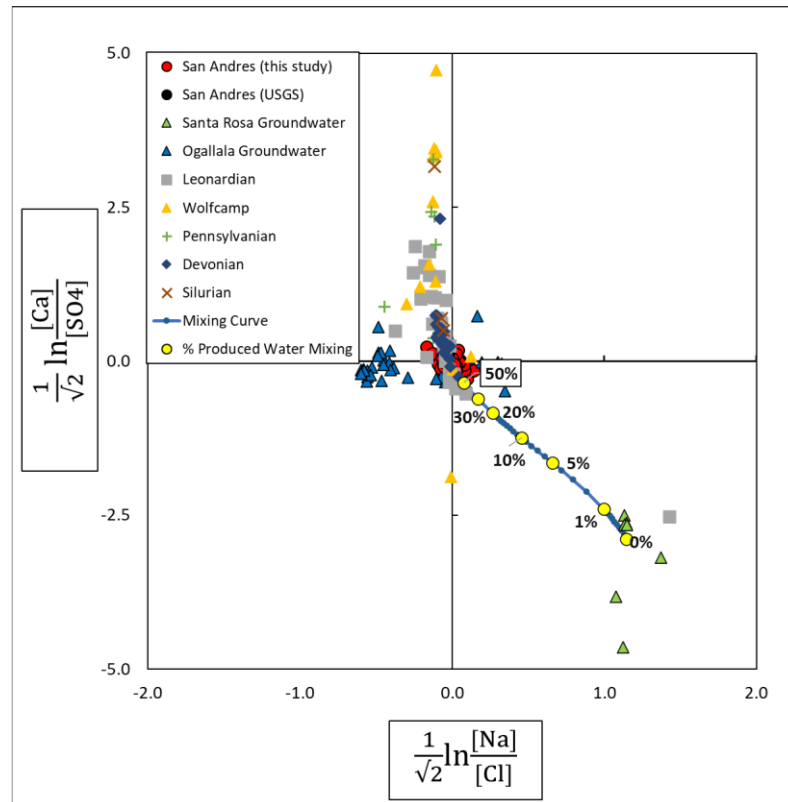
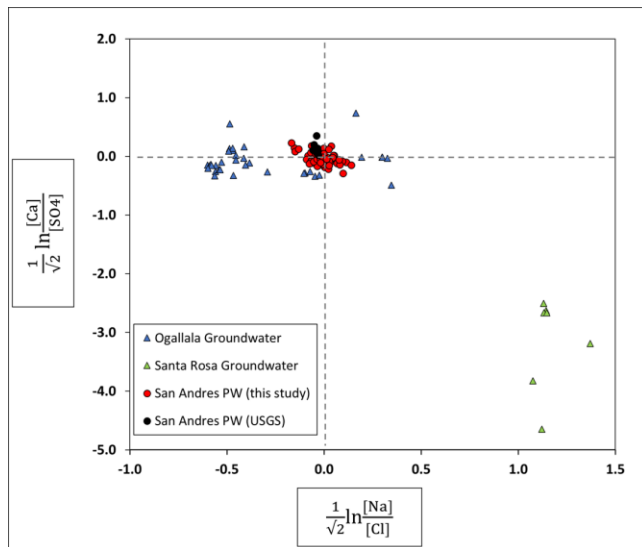
# Results: Groundwaters

- Ogallala and Santa Rosa Fm. groundwaters
  - Did not see significant change following CO<sub>2</sub> injection



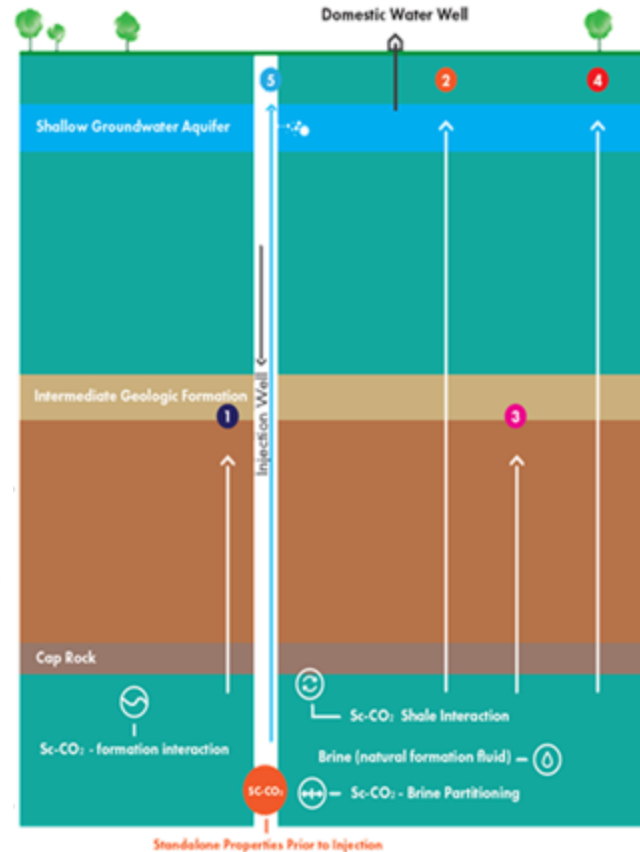
# Results: Groundwaters

- Ogallala and Santa Rosa Fm. groundwaters
  - Generated hypothetical mixing curve
    - Scenario: produced water migration into Santa Rosa Fm.



# Application of Field Results: GILD

- **Problem definition**
  - Need for a low-cost, easily implementable monitoring strategy for carbon storage reservoir leak detection
- **Proposed solution**
  - Geochemically Informed Leak Detection (GILD)



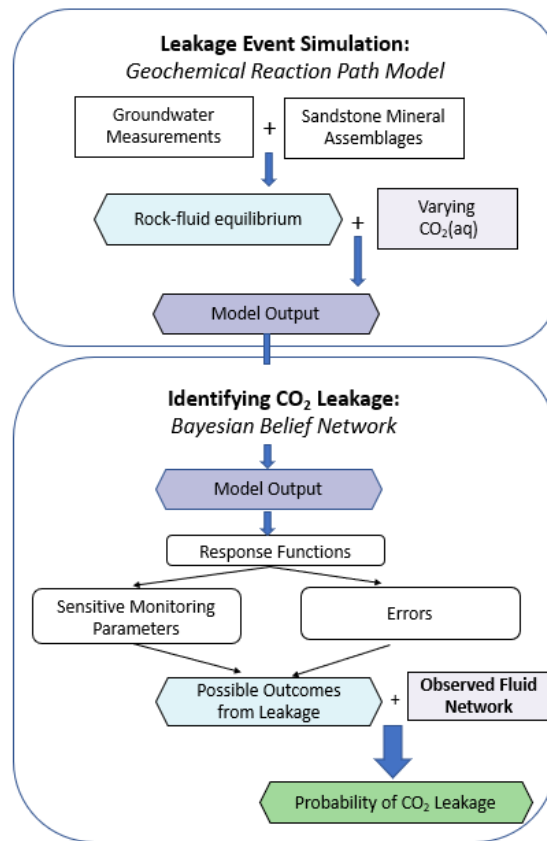
## Leakage Pathways

- 1 Wellbore to intermediate formation
- 2 Wellbore to shallow groundwater
- 3 Geologic conduit to intermediate formation
- 4 Geologic conduit to shallow groundwater
- 5 Well to shallow aquifer

# GILD Overview

## Geochemically Informed Leak Detection (GILD)

- 1) Assess fluid chemistry and mineral compositions of monitoring formation
- 2) Simulate leakage events with a geochemical reaction path model
- 3) Identify CO<sub>2</sub> leakage with a Bayesian Belief Network (BBN)

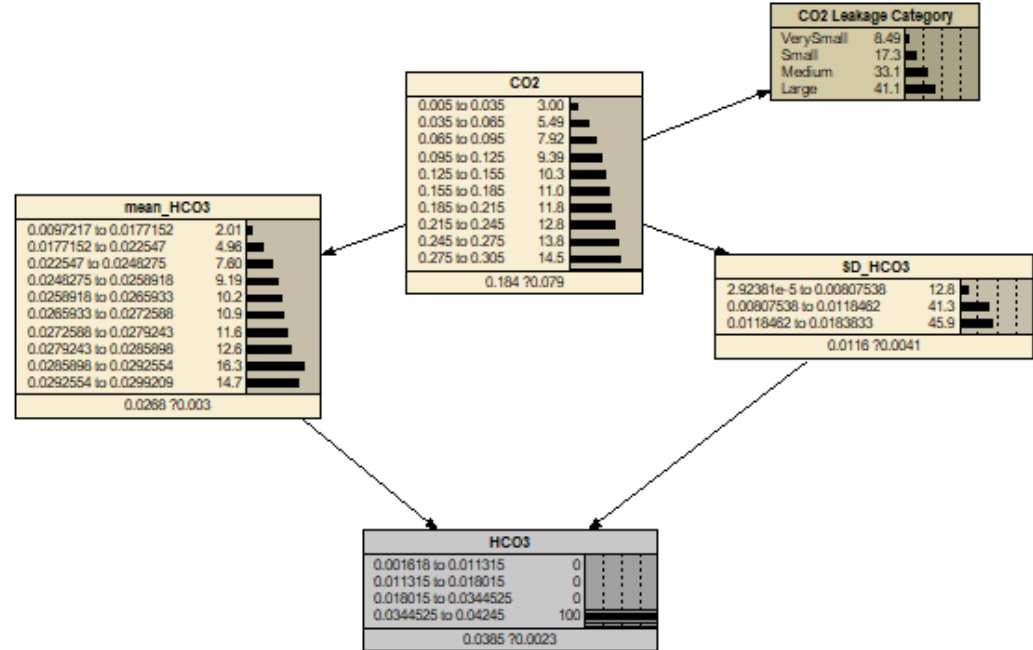


## Bayesian Belief Network (BBN)

- Decision support tool
- Probabilistic inference from multiple sources of evidence
- Application for leak detection: given monitoring parameters, compute the probabilities of the presence of leakage

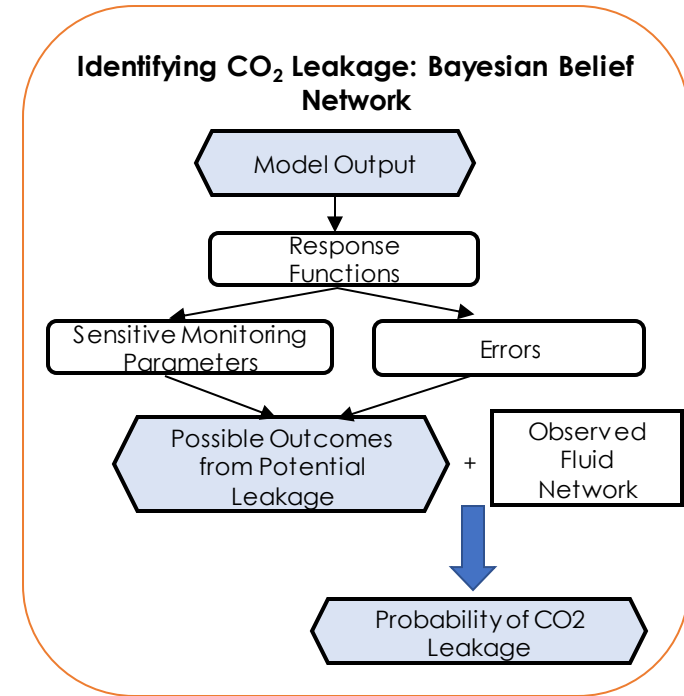
# Bayesian Belief Network (BBN) for Leak Detection

- **Upstream node**
  - CO<sub>2</sub> added concentration
- **Arrows**
  - Causal effect
- **Downstream nodes**
  - Monitoring parameters
- **Bars of each node**
  - Probability of a particular range
- **Conditional probability**
  - Probability of downstream given upstream
- **Purpose**
  - Back inference



# BBN Setup for Leak Detection

- **Observation parameters**
  - Inorganic geochemistry
    - $\text{Na}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ,  $\text{K}^+$ ,  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Br}^-$ , alkalinity as mg/L  $\text{HCO}_3^-$
- **Analytical techniques**
  - Fit response functions
  - Identify sensitive parameters
  - Include uncertainty
  - Predict possible outcomes from potential leakage
  - Compare with observed fluid network
  - Compute probability of  $\text{CO}_2$  leakage



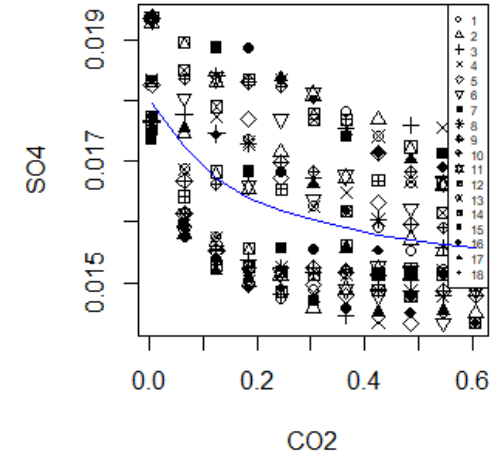
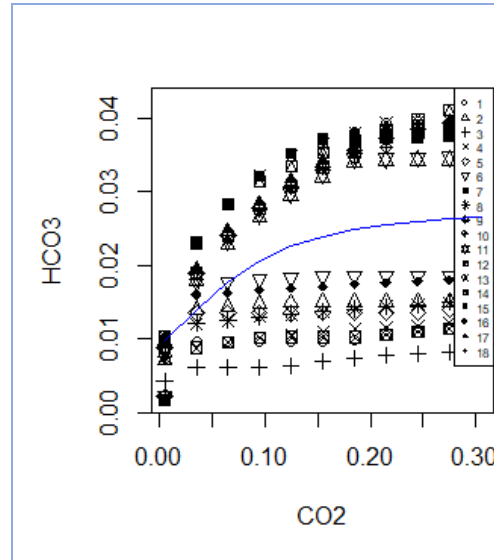
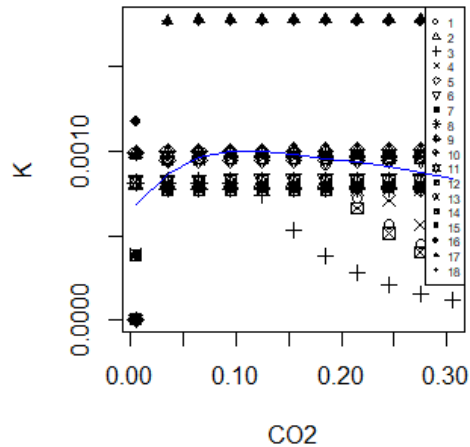
- Response functions

- Output of the geochemical model
- Relationship of ion concentrations (mol/kg) and CO<sub>2</sub> added concentrations (mol/kg)



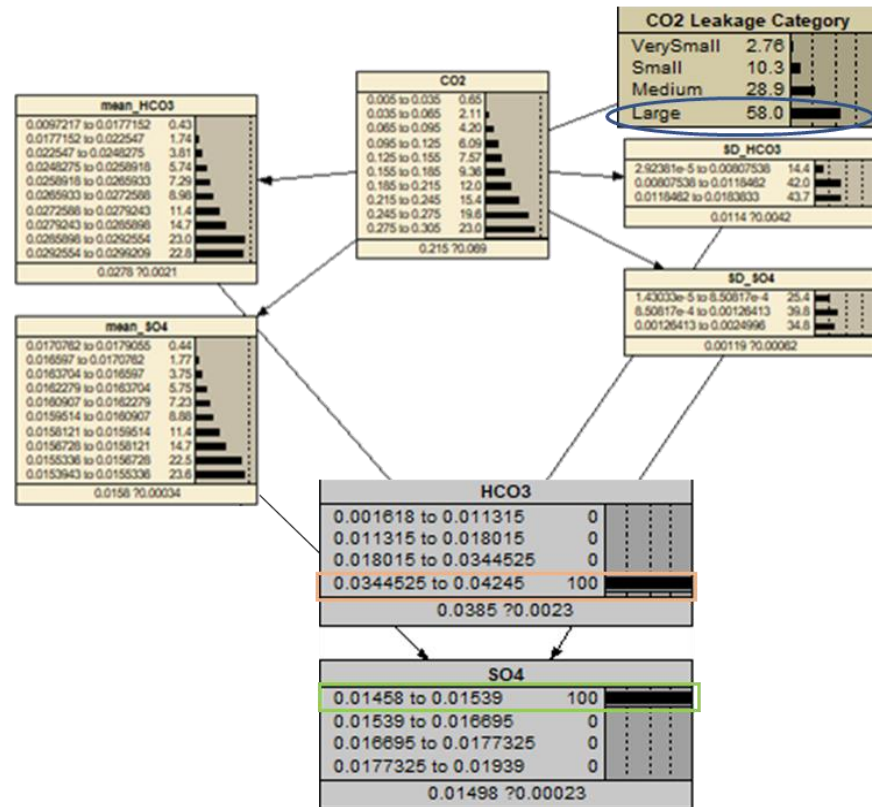
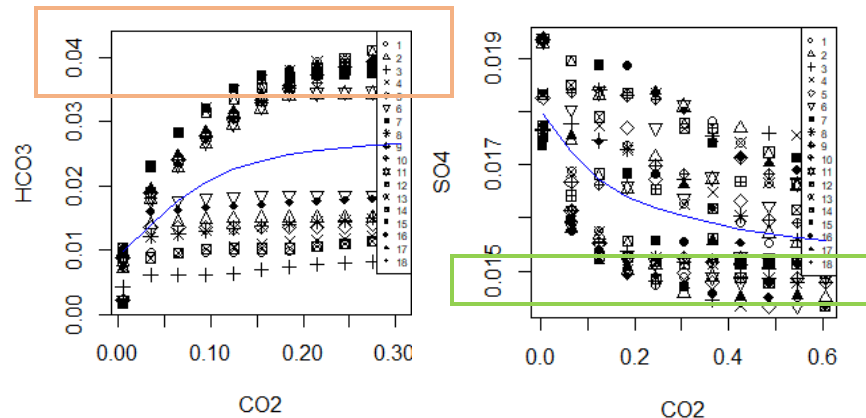
- Sensitive Parameters

- Selection Criterion
  - Monotonic relationship



# Application of BBN-Combining Evidence

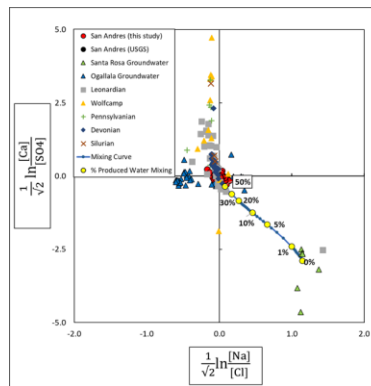
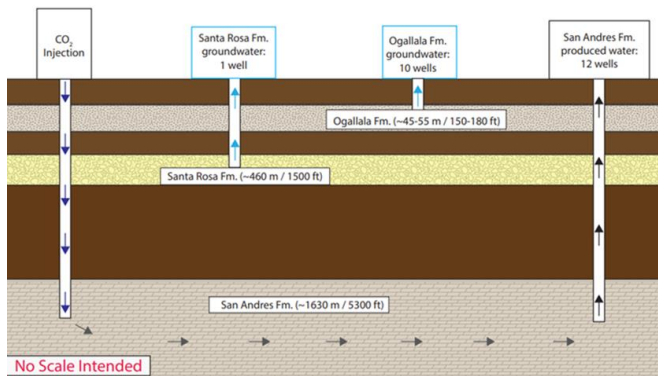
- Evidence from high  $\text{HCO}_3$  and low  $\text{SO}_4$





# Conclusions

- **Geochemical monitoring provides insight into groundwaters and target formation reactions**
  - No significant dissolution of storage reservoir
  - No produced water migration detected in groundwaters
- **Geochemical-statistical model (GILD) can provide CO<sub>2</sub> leakage detection via robust statistical analysis**
  - Current model applies user input via licensed software and researcher knowledge
  - Goal is to create standalone version that groundwater observers can use



## References

- Balch, R., McPherson, B. and Grigg, R., 2017. Overview of a large scale carbon capture, utilization, and storage demonstration project in an active oil field in Texas, USA. Energy Procedia, 114, pp.5874-5887.

## Publications from this Field Work

- Pfister, S., Capo, R.C., Stewart, B.W., Macpherson, G.L., Phan, T.T., Gardiner, J.B., Diehl, J.R., Lopano, C.L. and Hakala, J.A., 2017. Geochemical and lithium isotope tracking of dissolved solid sources in Permian Basin carbonate reservoir and overlying aquifer waters at an enhanced oil recovery site, northwest Texas, USA. Applied Geochemistry, 87, pp.122-135.
- Gardiner, J., Thomas, R.B., Phan, T.T., Stuckman, M., Wang, J., Small, M., Lopano, C. and Hakala, J.A., 2020. Utilization of produced water baseline as a groundwater monitoring tool at a CO<sub>2</sub>-EOR site in the Permian Basin, Texas, USA. Applied Geochemistry, 121, p.104688.
- Miller, J.D., Stuckman, M.Y., Means, N., Lopano, C. and Hakala, J.A., 2022. Determination of transition metal ions in fossil fuel associated wastewaters using chelation ion chromatography. Journal of Chromatography A, p.462924.

# Acknowledgement

- We would like to thank the operating company and its employees for helping us to sample the wells featured in this study.
- We would also like to thank the landowners who allowed us to sample their groundwater wells.

# NETL RESOURCES

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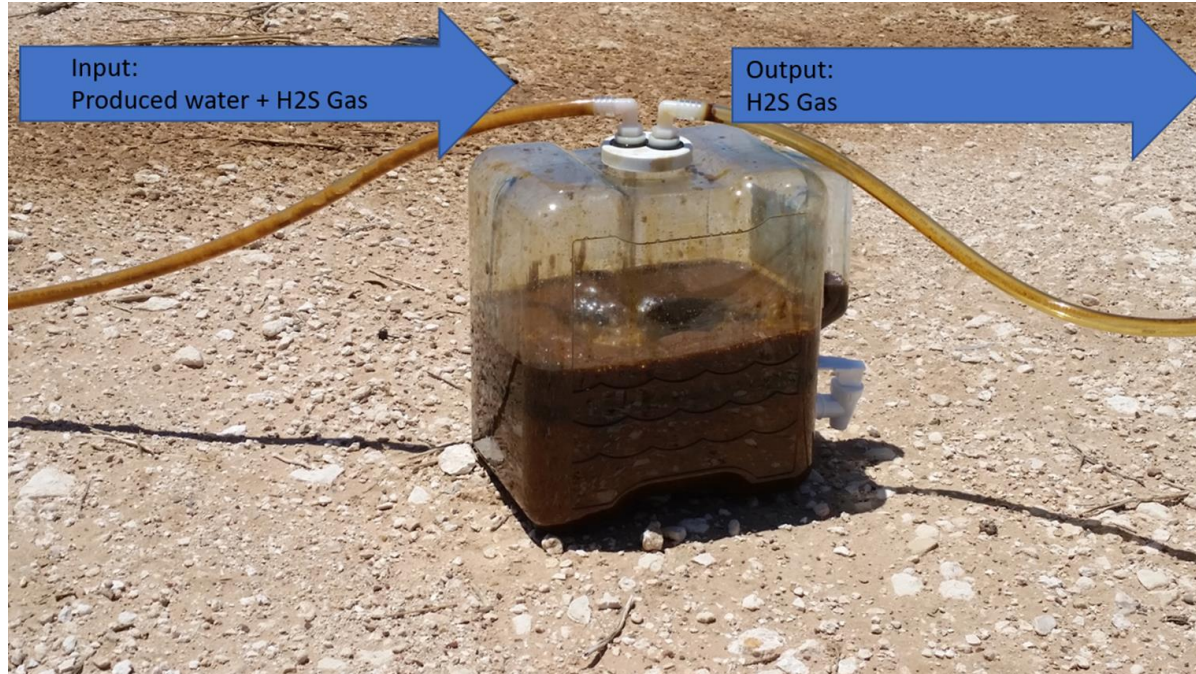
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# Supplemental Slides

# Field Sampling



Fluid from the well (e.g. oil, gas condensate, produced water, H<sub>2</sub>S gas) flows into the carboy and excess gas flows out.

# Construction of BBN-Regression Results

- Mean-predicted value from the response function for each CO<sub>2</sub> level
- SD-standard deviation of residuals (geochemical model output-predicted values)
- Ion concentration-normally distributed with mean and SD

